

Effect of Impurities on Supercritical CO₂ Compatibility

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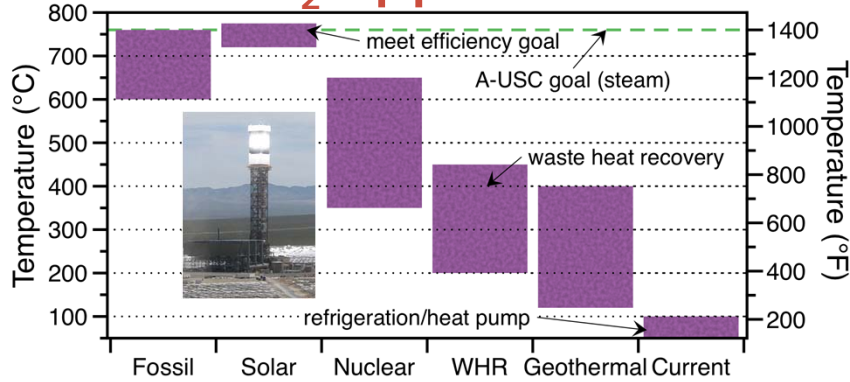
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Acknowledgments

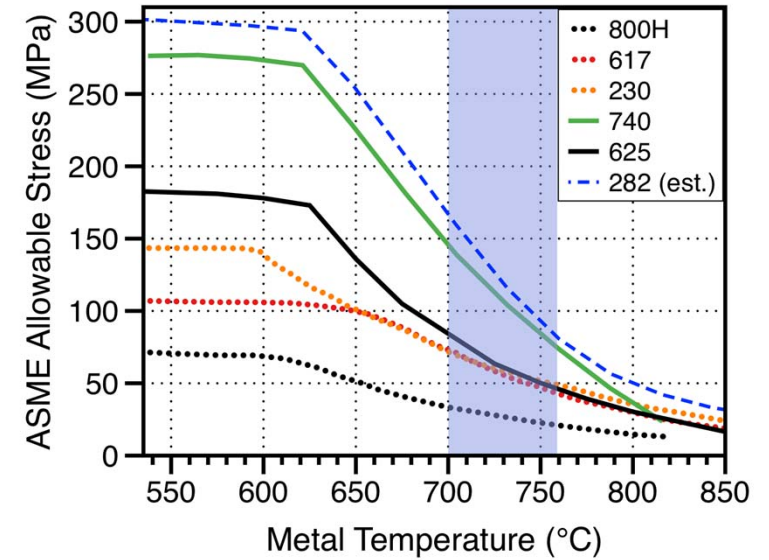
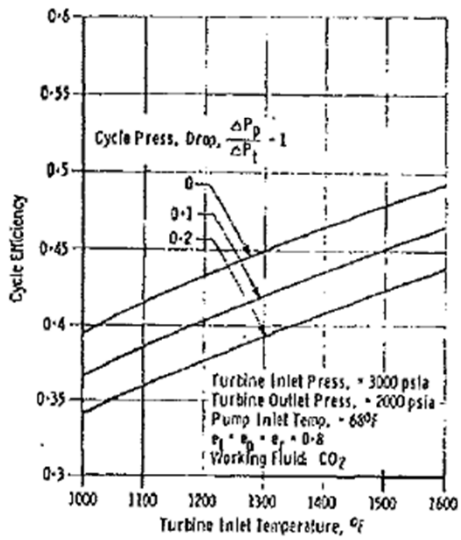
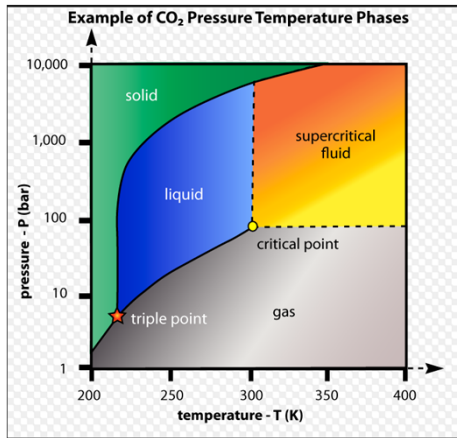
- J. Keiser, M. Howell, M. Stephens — oxidation experiments
- T. Lowe — SEM, image analysis
- T. Jordan — metallography
- D. Coffey — TEM specimen preparation
- Special thanks for alloys:
 - Haynes International
 - Special Metals
 - Sandvik
 - Capstone Turbine Corp.

Fossil/Solar focus on $>700^{\circ}\text{C}$ for high efficiency sCO_2

sCO_2 applications



$>700^{\circ}\text{C}$: favors precipitation-strengthened Ni-based alloys



740/282 above 700°C

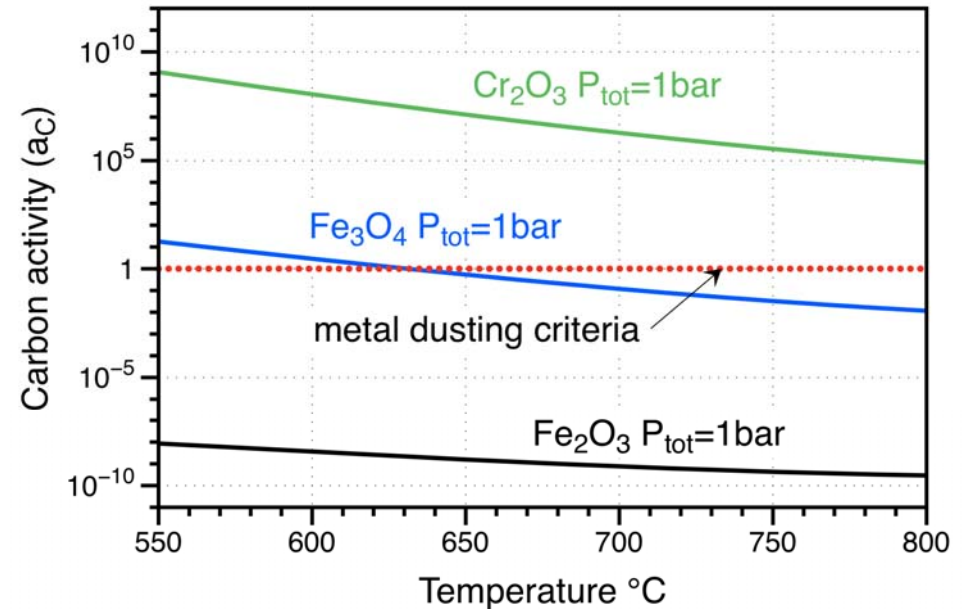
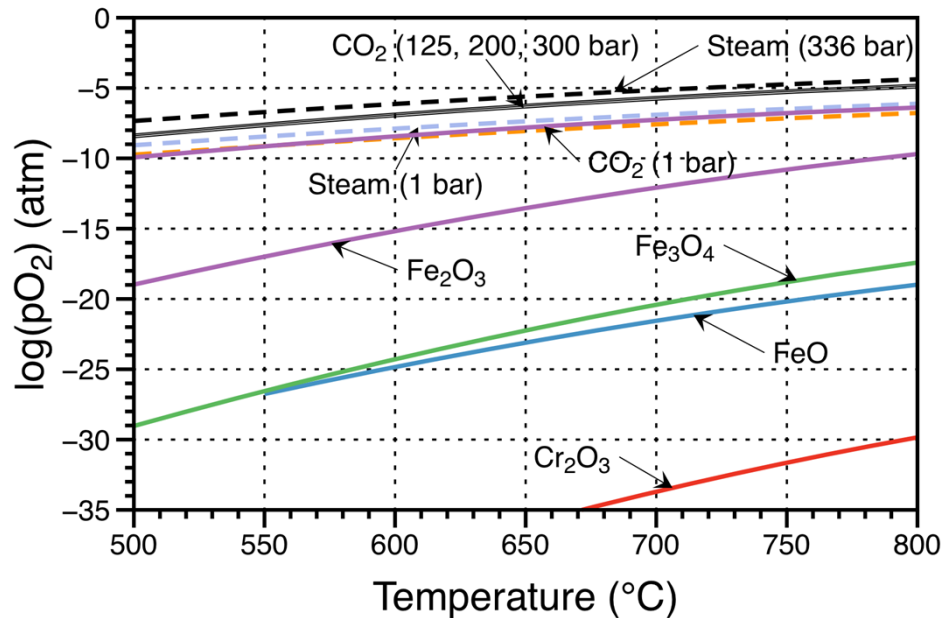
Feher, 1965

50% sCO_2 eff @ $>720^{\circ}\text{C}$

- Low critical point ($31^{\circ}\text{C}/7.4\text{ MPa}$)
- High, liquid-like density
- Flexible, small turbomachinery

Thermodynamics: Oxygen levels similar in steam/CO₂

Concern about high C activity at m-o interface



Factsage calculations



High carbon activity at $P_{total} = 1$ bar (What is $P_{interface}$?)

General conclusion: internal carburization not an issue for Ni-based alloys

Supercritical CO₂ Allam cycle: first clean fossil energy?

NetPower 25MWe demo plant (Texas)

Exelon, Toshiba, CB&I, 8Rivers Capital: \$140m



The prototype NET Power plant near Houston, Texas, is testing an emission-free technology designed to compete with conventional fossil power.

CHICAGO BRIDGE & IRON

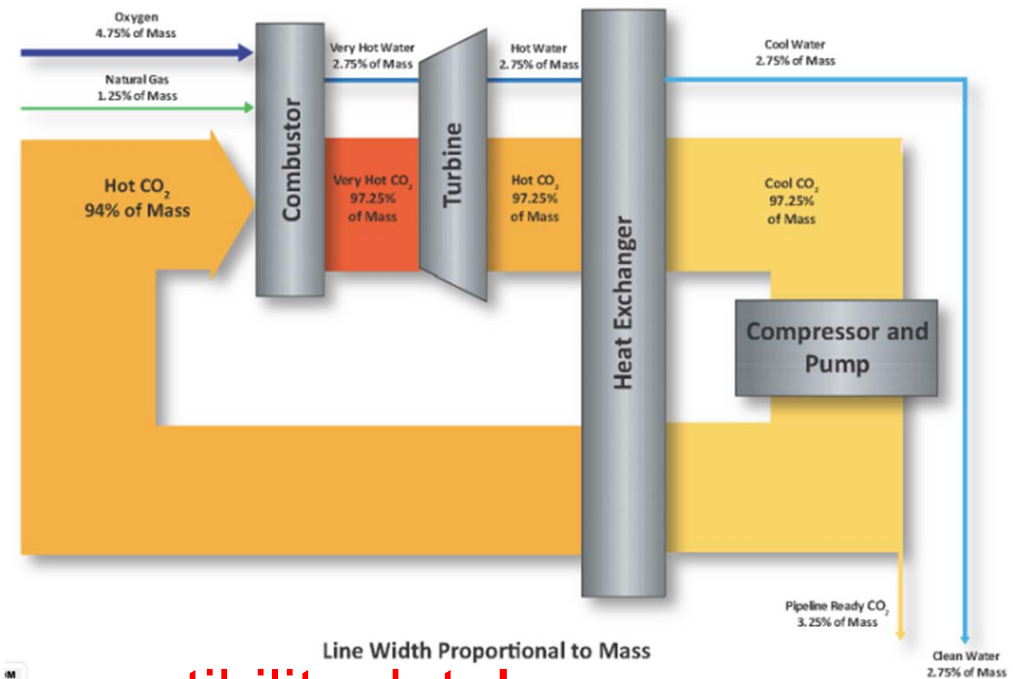
May 2018: announced first firing

Material challenges:

Combustor: 1150°C (!?!)

Turbine exit: 750°C/300 bar

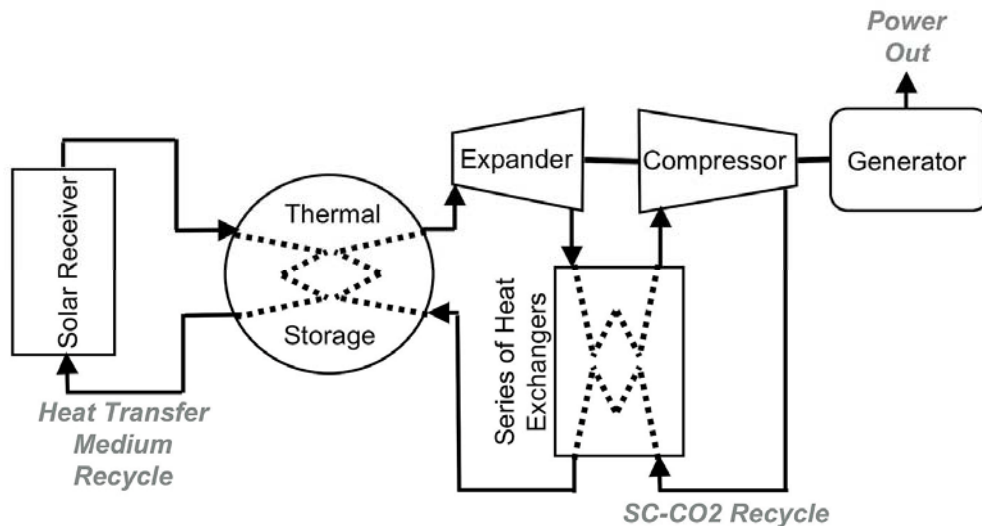
Combustion impurities: O₂, H₂O, SO₂



**Moving forward with limited compatibility data!
As audacious as Eddystone in 1960**

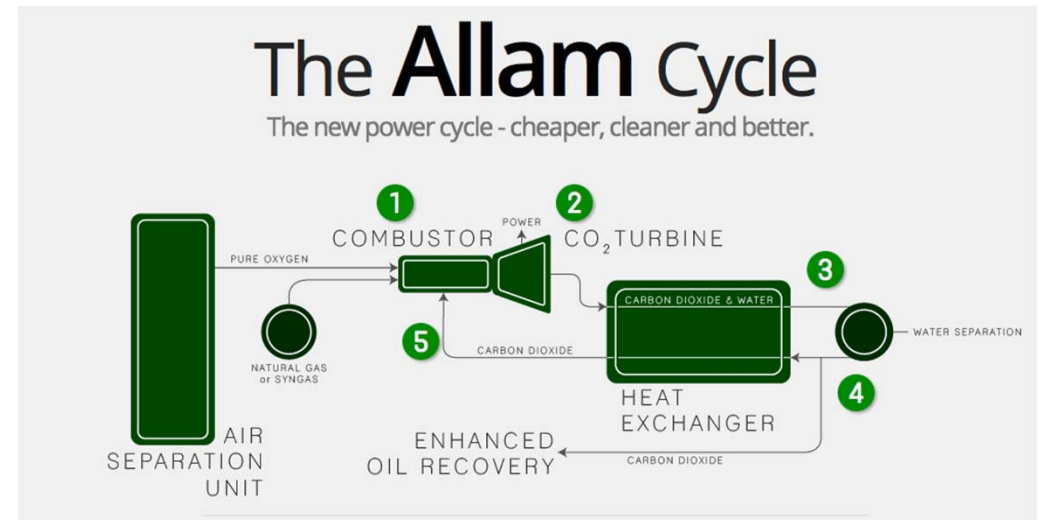
Impurities differ in indirect- & direct-fired sCO₂ cycles (i.e. closed vs. open)

Closed cycle:
"pure" CO₂ 100-300 bar



DOE SunShot 3 yr. funding
Impurities: lower cost, industrial grade CO₂

Open cycle:
sCO₂ + impurities (O₂, H₂O...)



DOE Fossil Energy 3 yr. funding
Impurities: high levels from combustion residue

Two sCO₂ projects completed at ORNL

DOE Fossil Energy (2015-2018)

- 750°C/300 bar: 500-h cycles
- Focus on impurity effects for direct-fire
 - Baseline research grade (RG) CO₂
 - New autoclave with controlled O₂+H₂O
- Alloys
 - 310HCbN (HR3C, Fe-base SS)
 - 617
 - 230
 - 247 (Al₂O₃-forming superalloy)
 - 282 (Heat #2)
 - 740

DOE SunShot (CSP) (2015-2018)

- 750°C/300 bar: 500-h cycles
 - Including 750°C/1 bar, 10-h cycles
- Focus on industrial grade (IG) CO₂
 - Indirect fired (closed loop)
- Alloys
 - Alloy 25 (Fe-base SS Sanicro 25)
 - 625
 - 740
 - 282 (Heat #1)

Combined test matrix:

	FE/CSP	FE	CSP	FE	CSP
	Air	RG CO ₂	IG CO ₂	CO ₂ +O ₂ /H ₂ O	CO ₂ +50/50 ppm
1 bar	5,000 h	5,000 h	10,000 h	—	—
300 bar	—	5,000 h	10,000 h	5,000 h	2,500 h

Range of alloys have been evaluated

	Ni	Cr	Fe	Co	Refractories	Ti	Al	S	Other
Grade 91	0.1	8.3	90	0.01	0.9Mo,0.1Nb	<0.01	<0.01	10	0.03Cu,0.3Mn,0.1Si,0.3V
304H	8.4	18	70	0.1	0.3Mo,0.01Nb	<0.01	<0.01	29	0.4Cu,1.6Mn,0.3Si,0.07N
25	25	22	43	1.5	3.5W,.5Nb,.2Mo	0.02	0.03	8	3.0Cu, 0.5Mn, 0.2Si, 0.2N
310HCbN	20	25	51	0.3	0.1Mo,0.4Nb	0.01	<0.01	<10	0.1Cu,1.2Mn,0.3Si,0.3N
230	61	23	2	0.1	1Mo, 12W	0.01	0.3	9	0.02La
625	61	22	4	0.1	9Mo, 4Nb	0.2	0.1	<10	0.2Si,0.1Mn,0.02C
617	54	22	1	13	9Mo, 1Nb	0.3	1.1	<3	
740	48	23	2	20	0.3Mo, 2Nb	2.0	0.8	<10	0.5Si,0.3Mn,0.03C
282	58	19	0.2	10	8Mo	2.2	1.5	<1	0.1Si,0.1Mn,0.06C
247	60	8	0.03	10	10W,3Ta,1Mo	1.1	5.3	<1	1.3Hf,0.14C

Compositions measured using ICP-OES and combustion analyses

CO₂ compatibility evaluated several ways at 700°-800°C

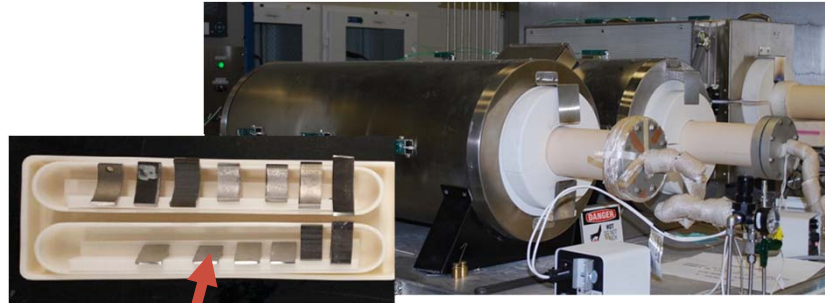
Autoclave: 300 bar sCO₂
500-h cycles



Correct temperature and pressure

4-5 cm² alloy coupons

Tube furnace: 1 bar CO₂
500-h cycles



Same cycle frequency as autoclave



Baseline:
Box furnace:
Lab. Air
500-h cycles

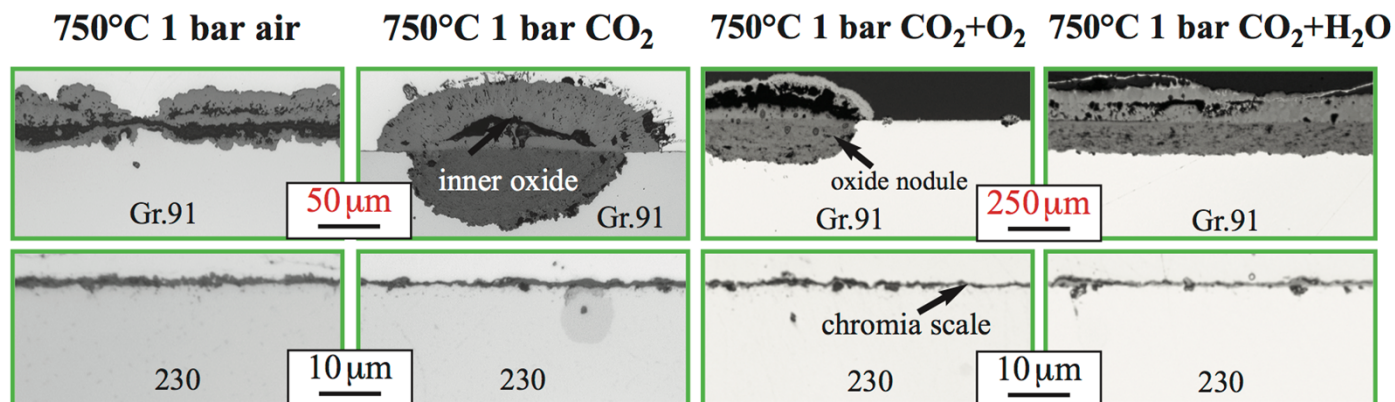
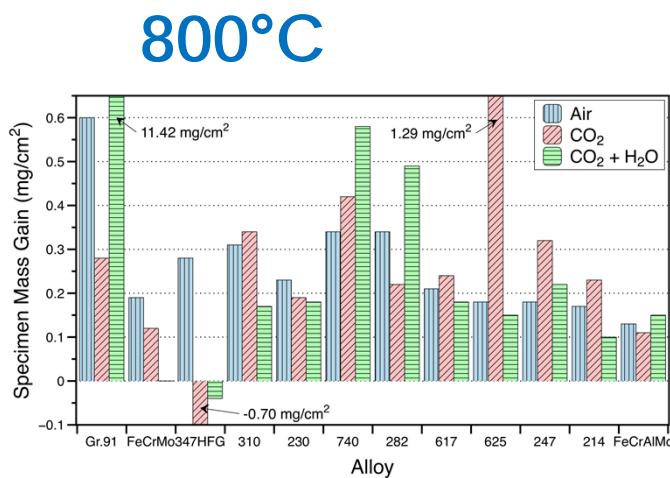
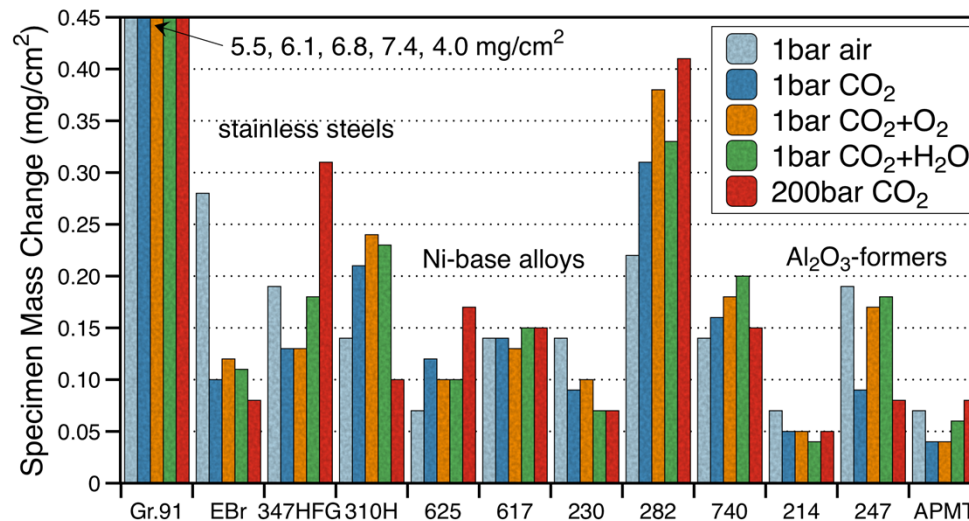
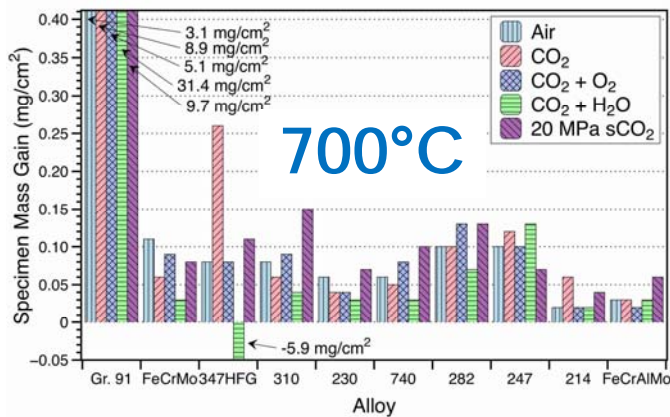
Previously:
"Keiser" rig:
500-h cycles, 1-43 bar CO₂



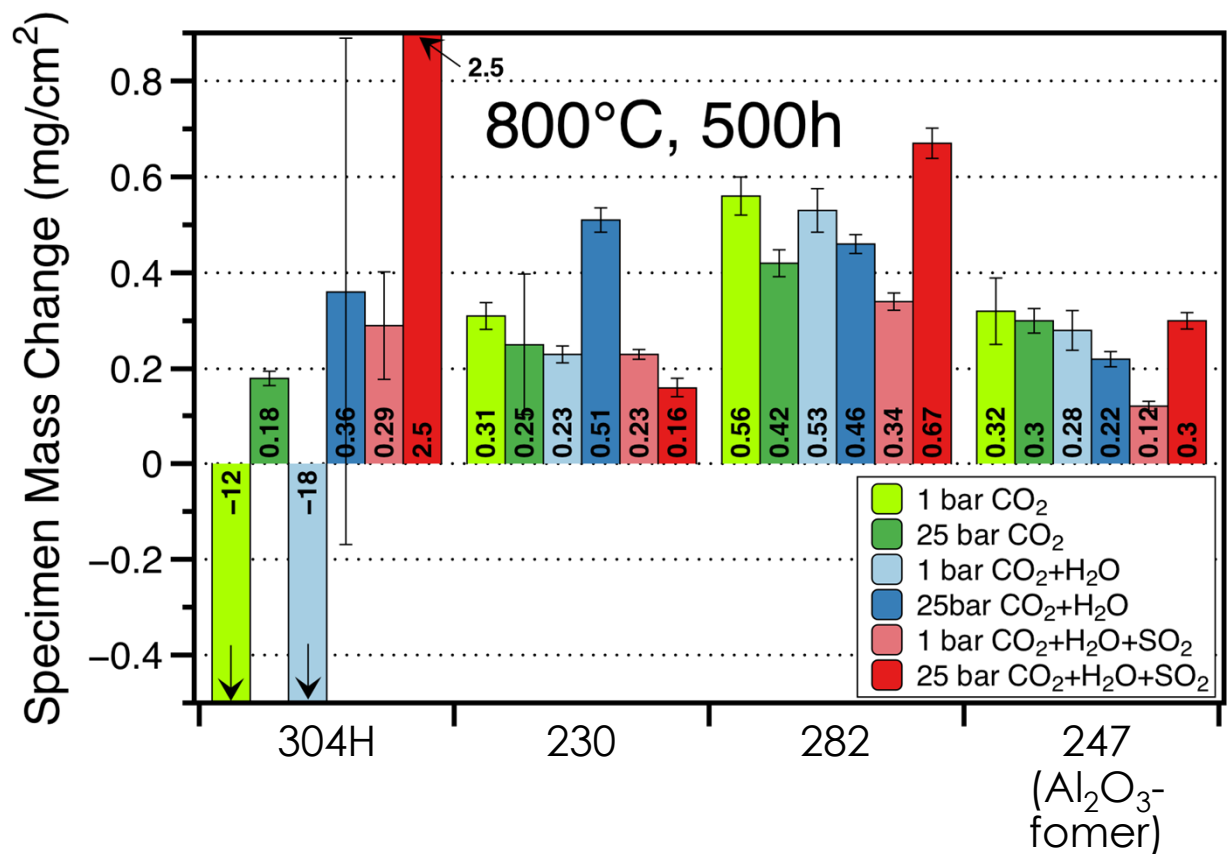
Study impurities at 1-43 bar

Baseline of research grade (RG) CO₂: ≤ 5 ppm H₂O and ≤ 5 ppm O₂
industrial grade (IG) CO₂: 18 ± 16 ppm H₂O and ≤ 32 ppm O₂

Impurities (2015): 1atm, many alloys (1 of each)



Impurities (2017): fewer alloys (3 of each), 1 and 25 bar



Two alloy 230 reaction tubes:

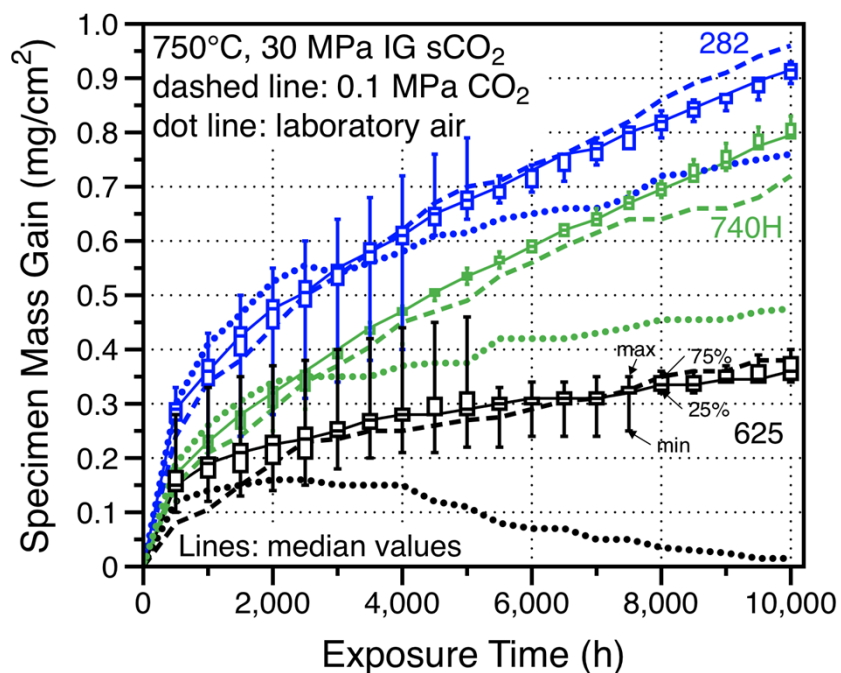
Pressure: 1 and 25 bar

Gas: RG CO₂
 CO₂+10%H₂O
 CO₂+10%H₂O+0.1%SO₂

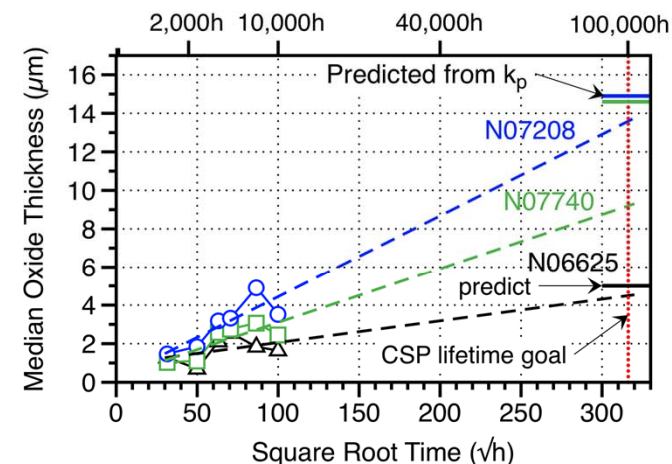
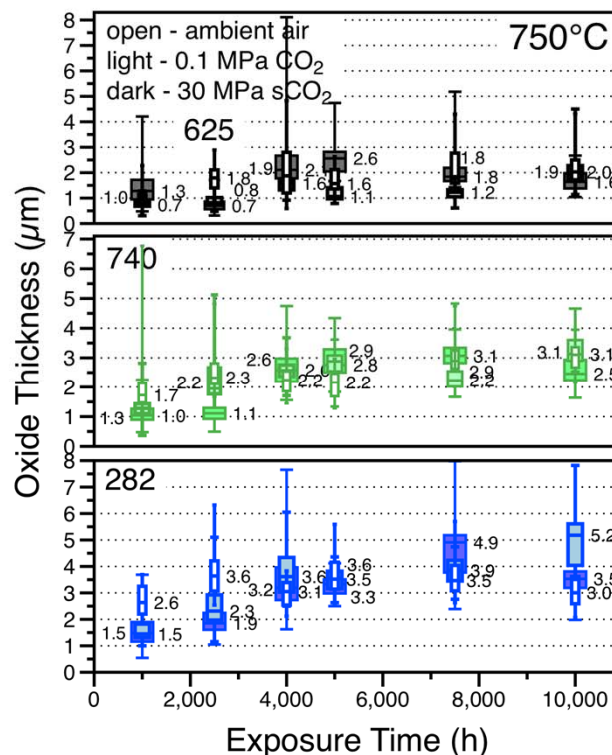
Pint, Brese, Keiser, NACE Corrosion 2018 C2018-11199

CSP: completed 10,000 h exposures for lifetime model

500-h cycles: three different environments



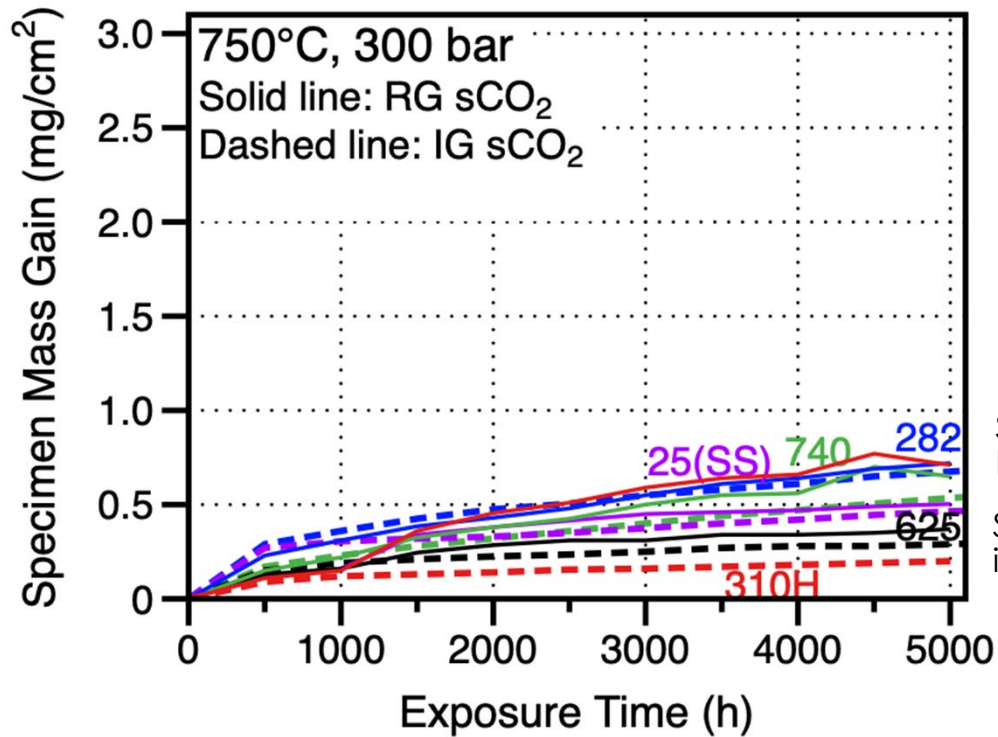
Quantification of scale thickness in three environments:



Extrapolation sCO₂ to 100,000 h

Pint, Keiser, NACE Corrosion 2019 C2019-12750

2016-2017: baseline performance in RG and IG sCO₂



Sanicro 25
Fe-22Cr-25Ni

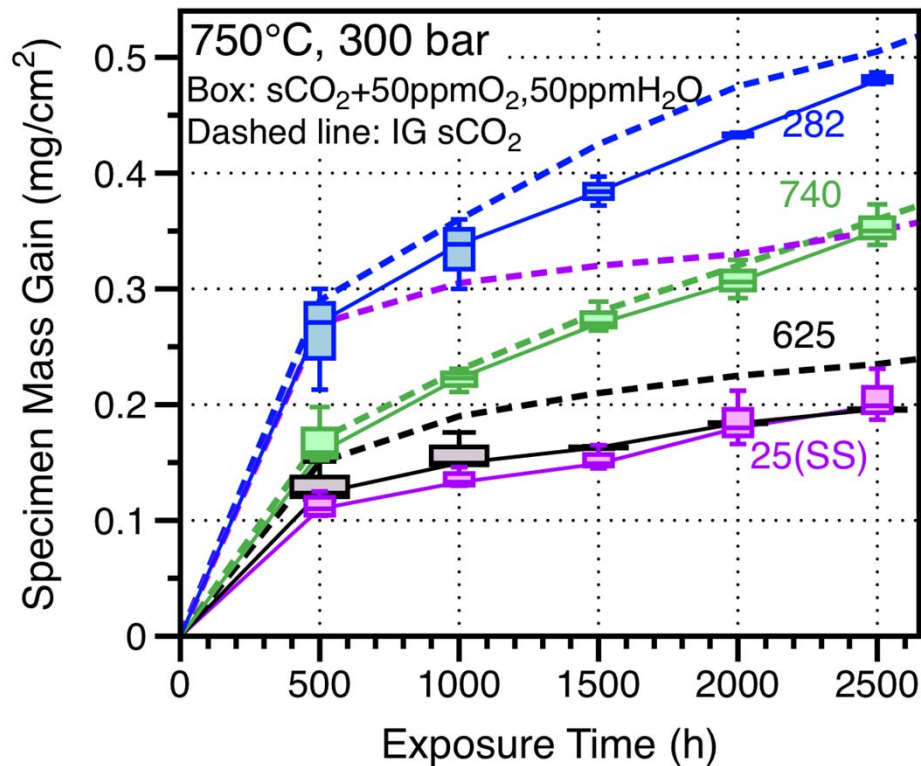
Similar mass gain
in RG & IG sCO₂



Research grade (RG) CO₂: ≤ 5 ppm H₂O and ≤ 5 ppm O₂
Industrial grade (IG) CO₂: 18±16 ppm H₂O and ≤ 32 ppm O₂

2018: finally completed multi-pump 300 bar autoclave

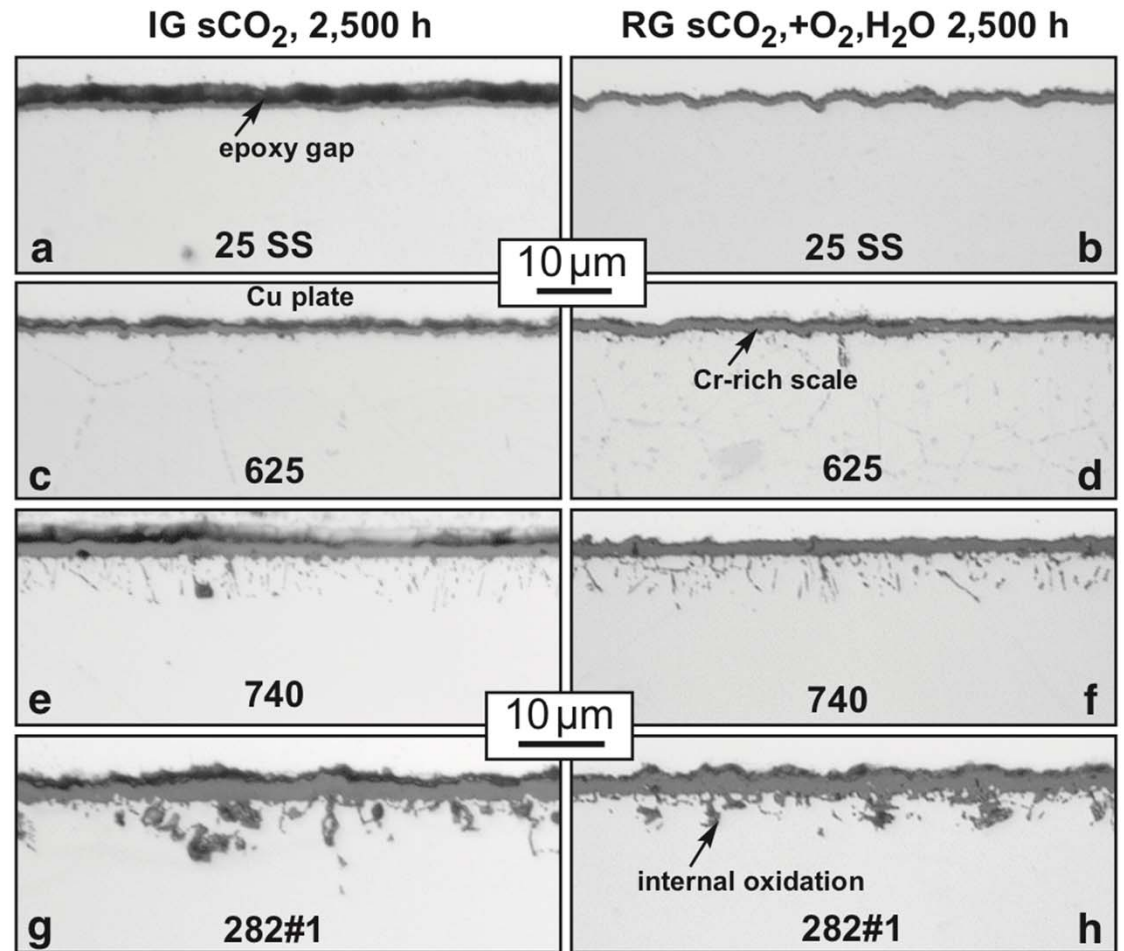
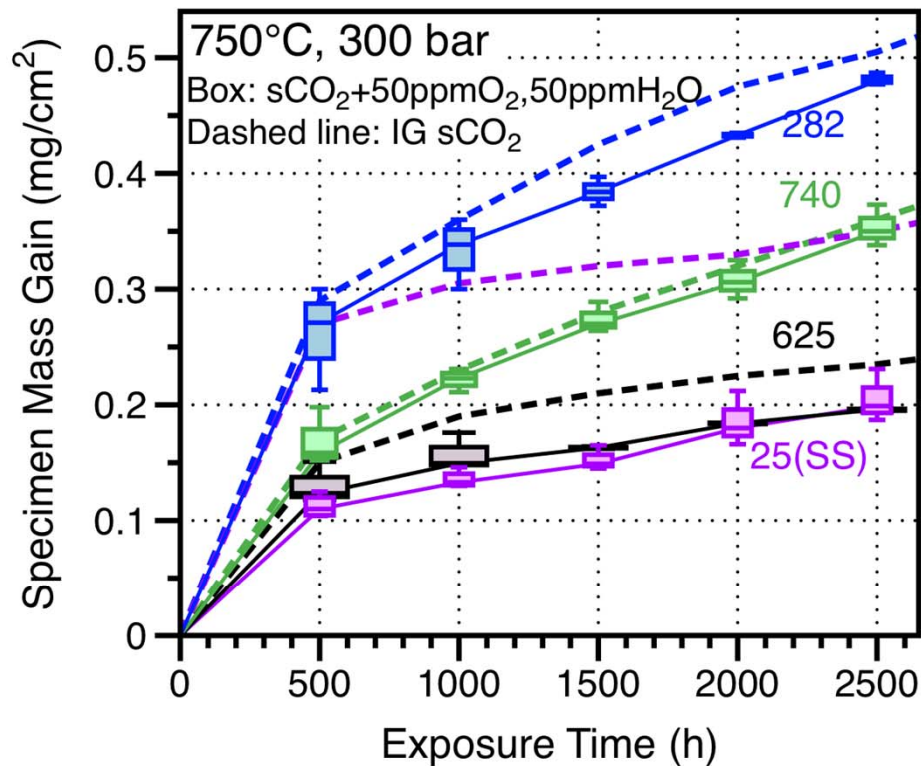
Closed cycle: low impurity levels showed no effect



- Box: 3-4 specimens
 - Whiskers: min/max values
 - One specimen removed at 1 kh
- Lines: median values
 - 8-11 specimens in IG sCO₂
- Minor changes between:
 - IG sCO₂ (18±16 ppm H₂O, ≤ 32 ppm O₂)
 - RG sCO₂+50ppmO₂+50ppmH₂O
- Transient effect for Fe-based 25

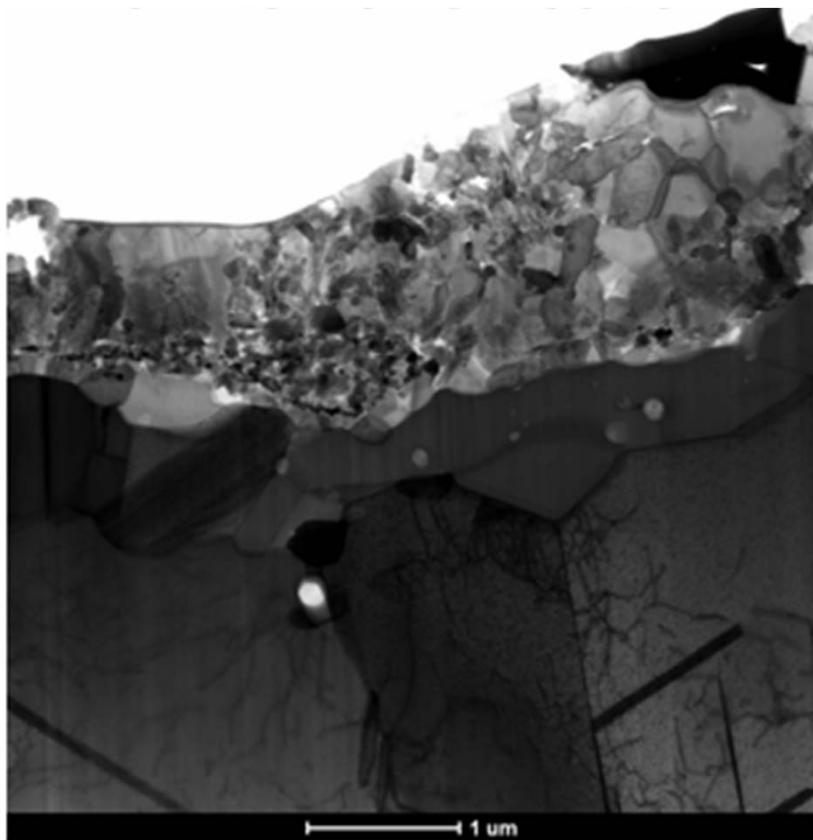
2018: finally completed multi-pump 300 bar autoclave

Closed cycle: low impurity levels showed no effect

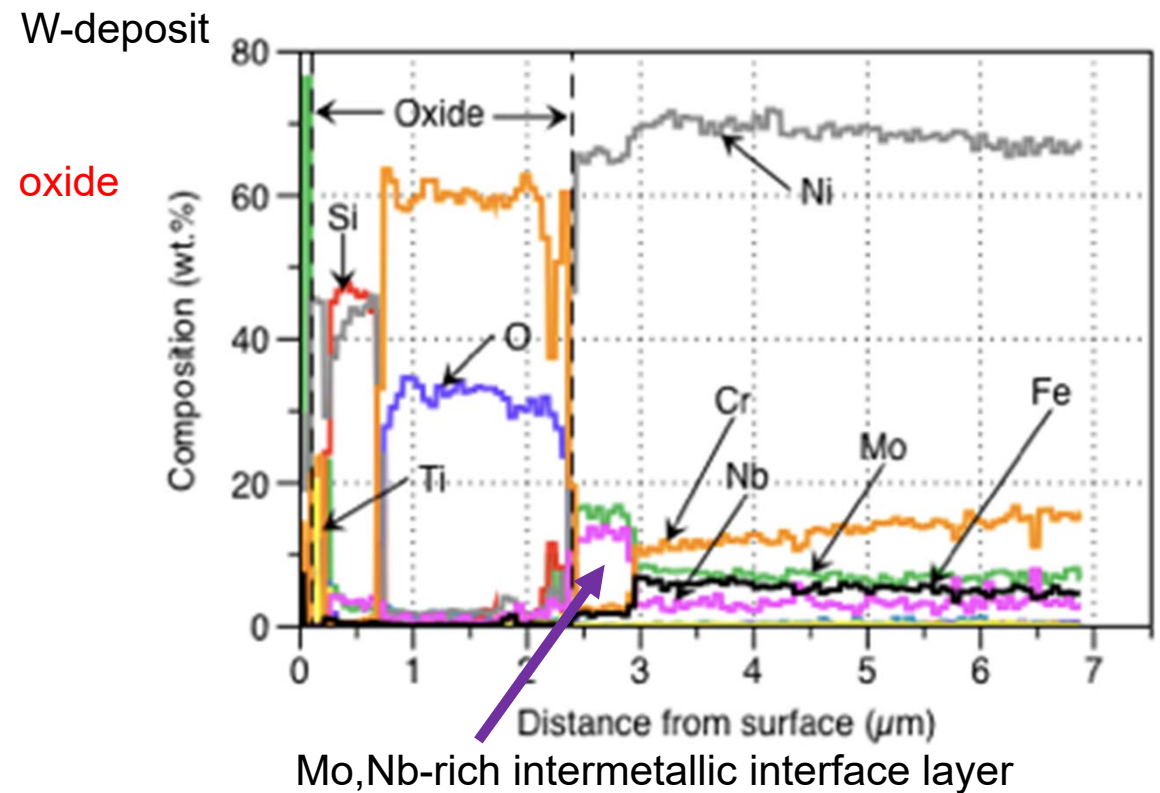


625: less complex reaction product than PS alloys
1,000 h at 750°C in 300 bar IG sCO₂

STEM BF image

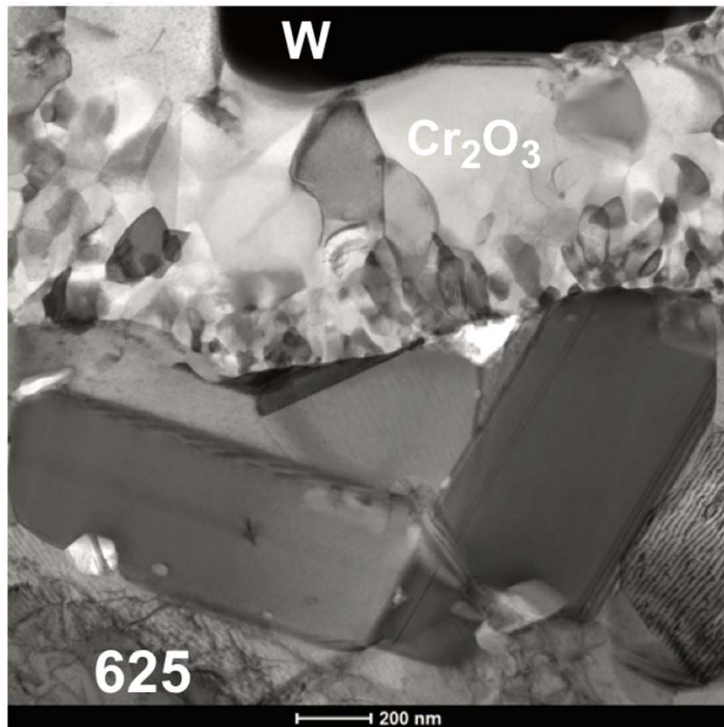


EDS line maps

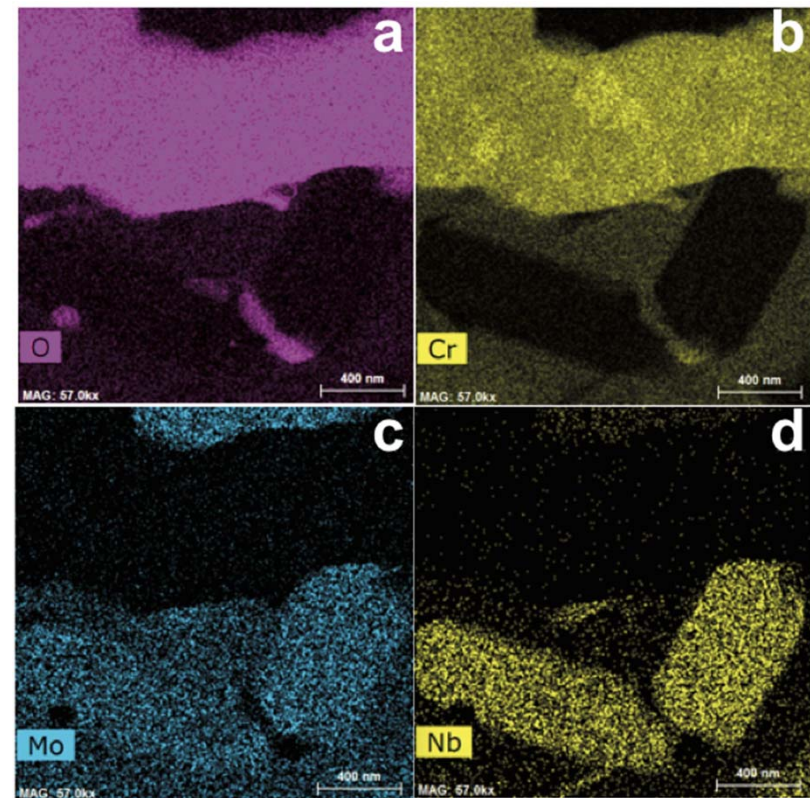


625: 1kh at 750°C in RG sCO₂+50ppmO₂/H₂O
- similar oxide as IG sCO₂, consistent with mass change

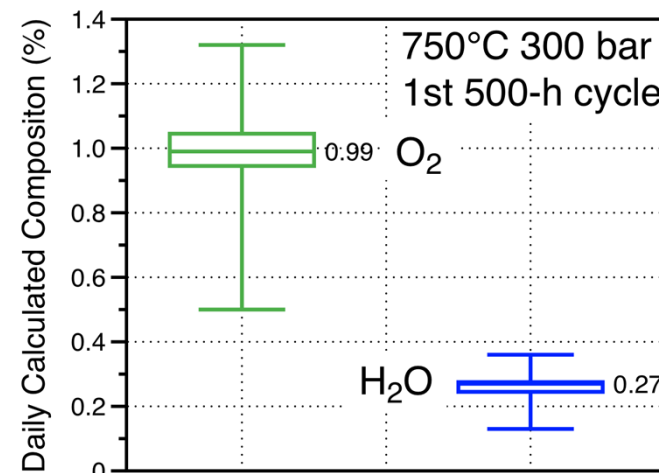
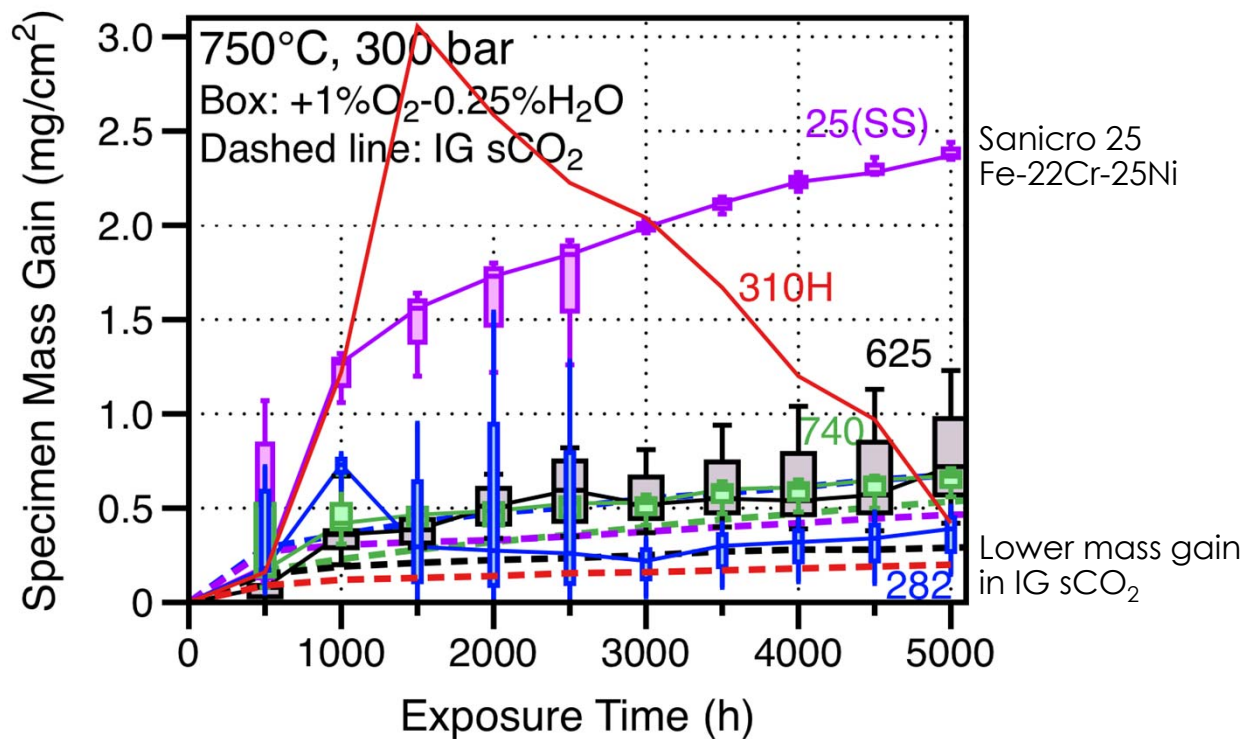
STEM BF image



EDS maps



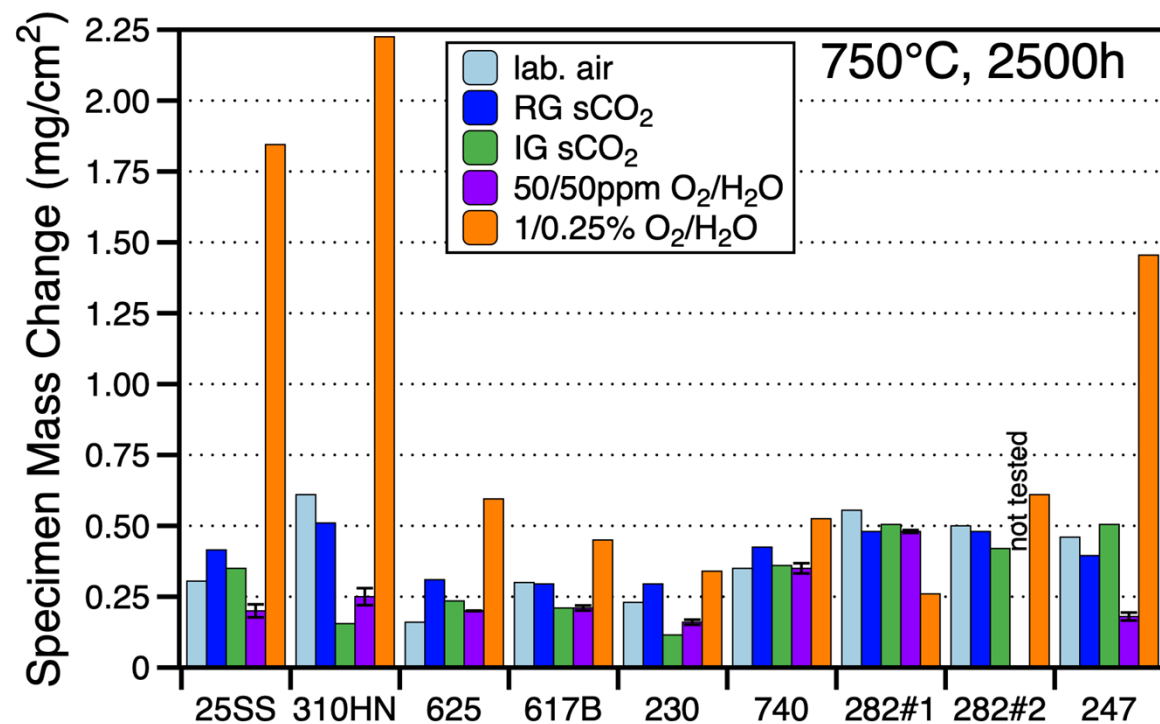
Open cycle: clearly see an effect of higher impurities



Goal: 1%O₂+0.25%H₂O
 (industry suggestion)
 Not easy to control at 300 bar

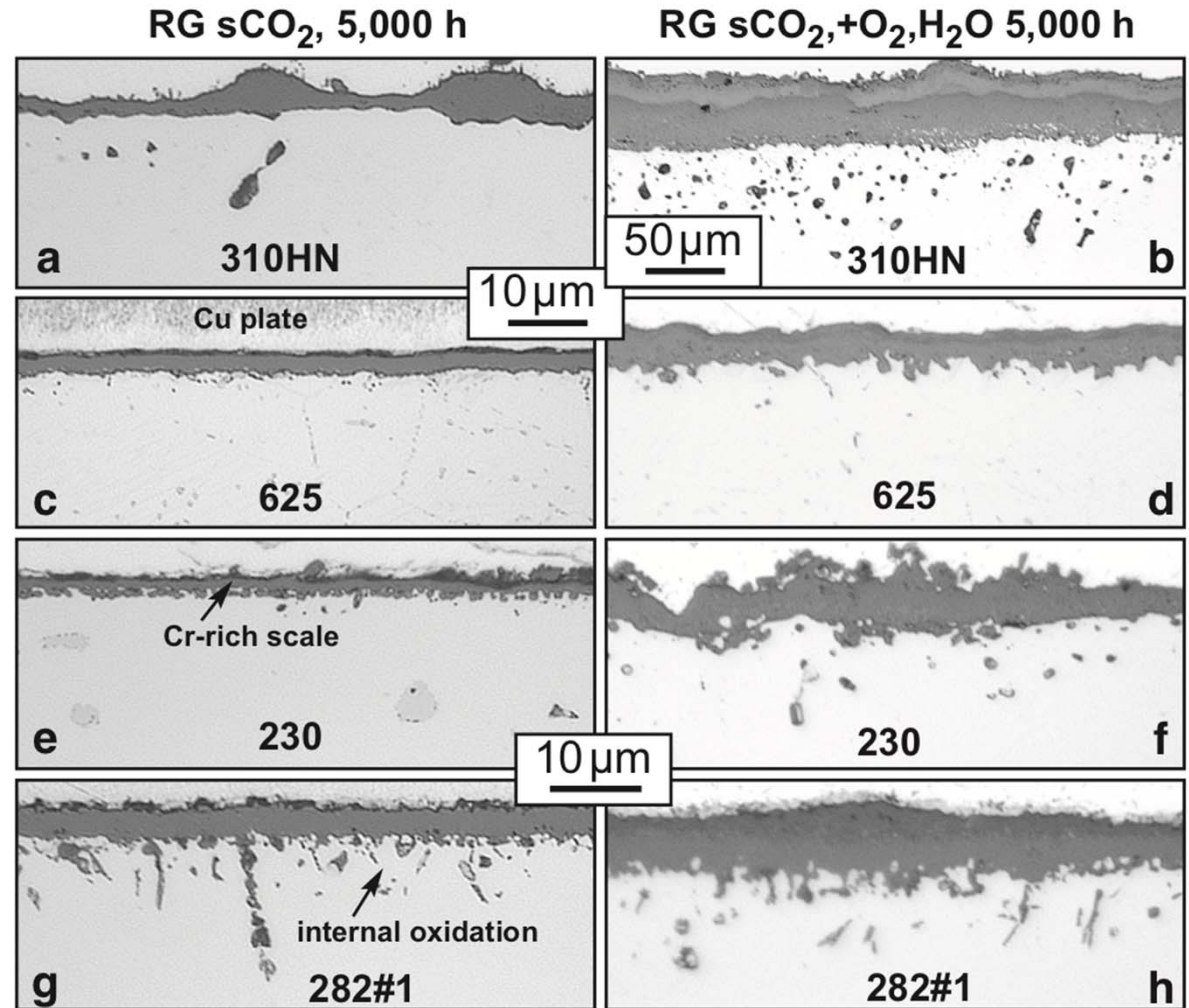
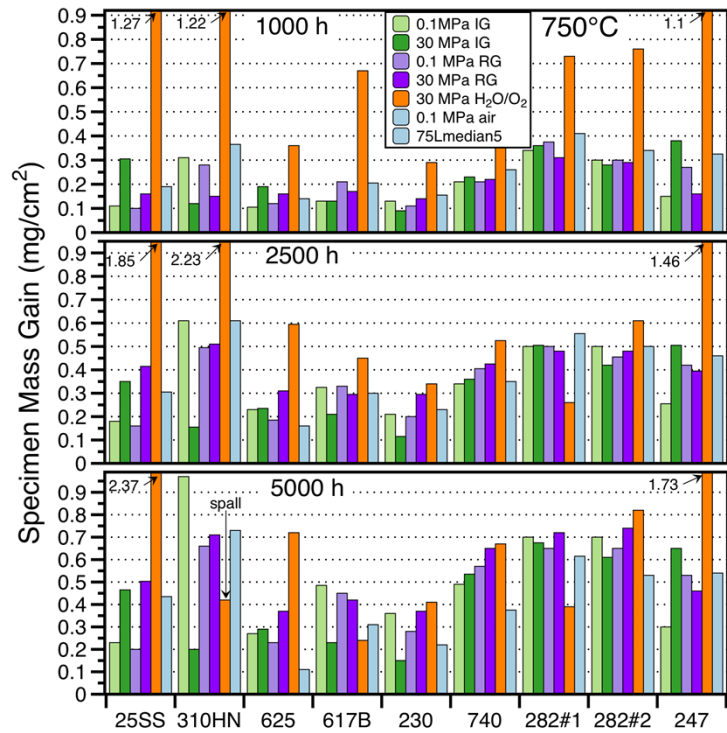
Industrial grade (IG) CO₂: 18±16 ppm H₂O and ≤ 32 ppm O₂

Summary: high impurities resulted in higher mass gains

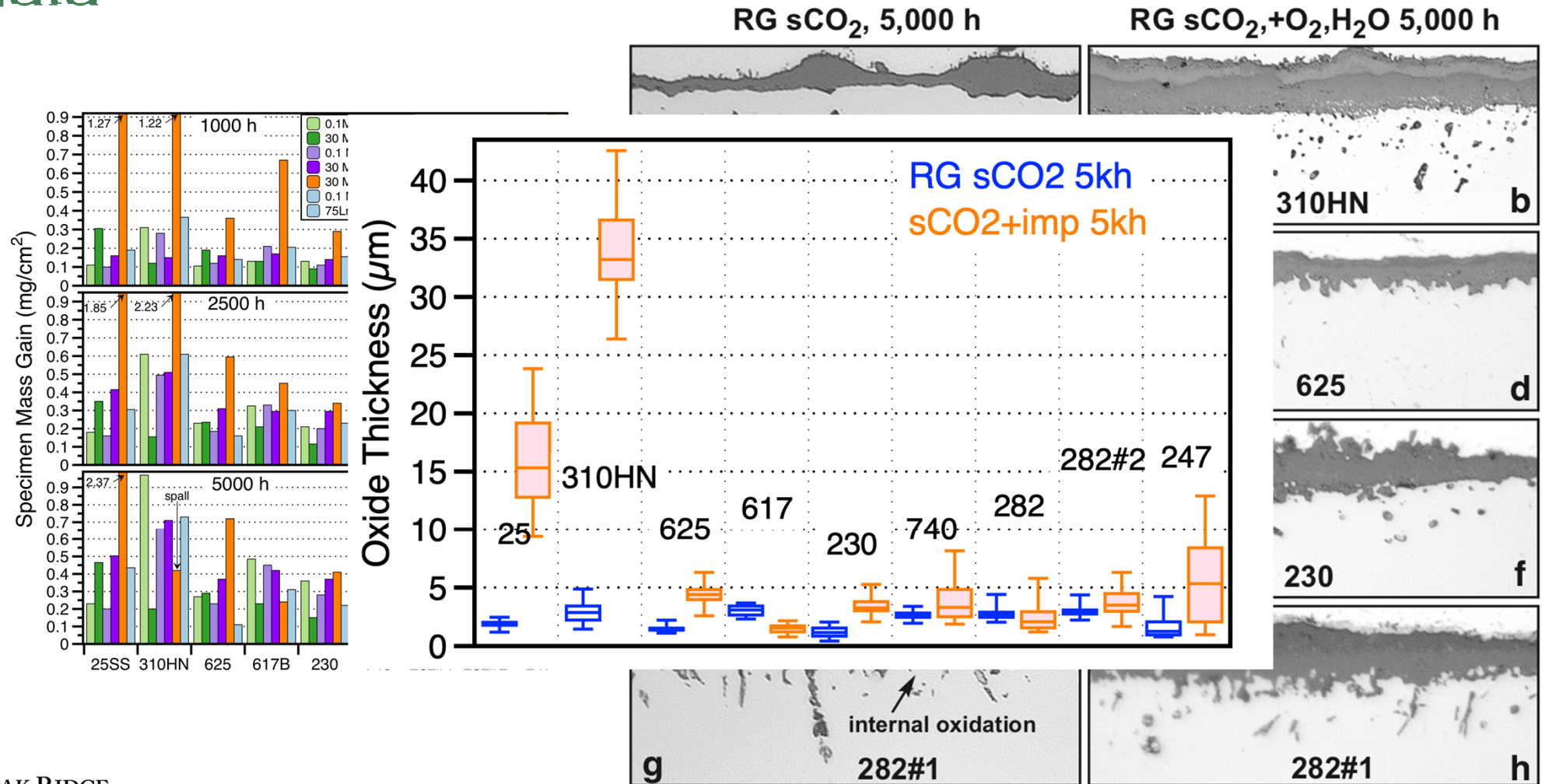


RG CO₂ + 1%O₂ + 0.25% H₂O

Metallography generally consistent with mass change data

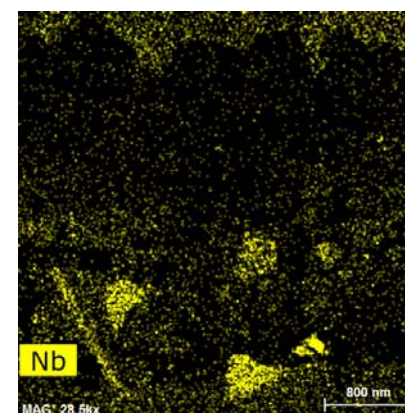
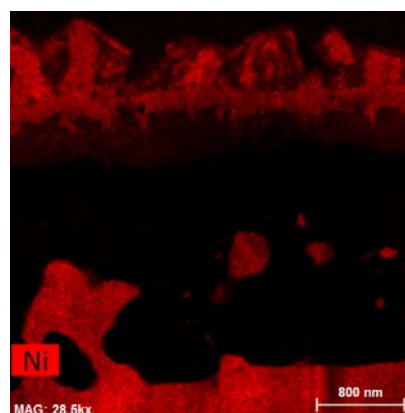
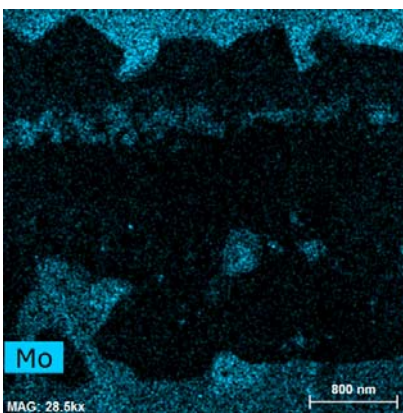
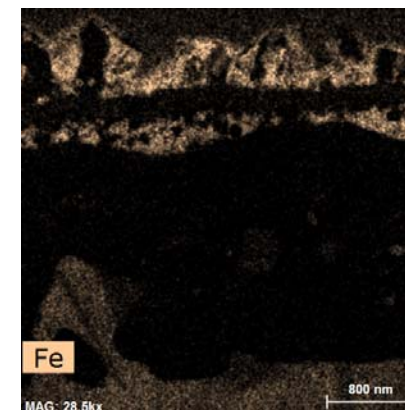
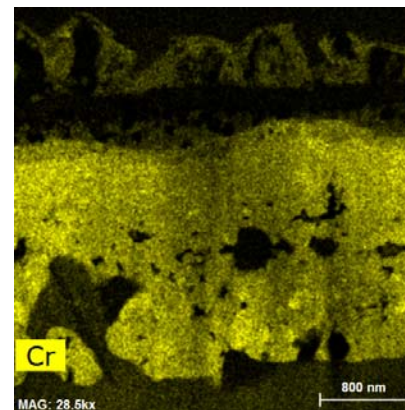
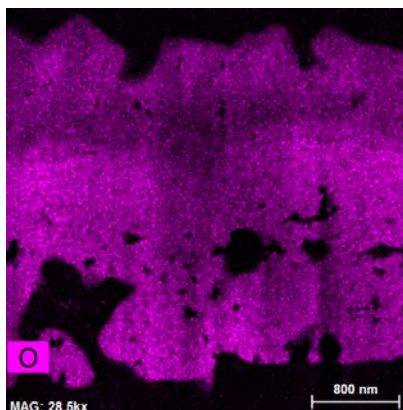
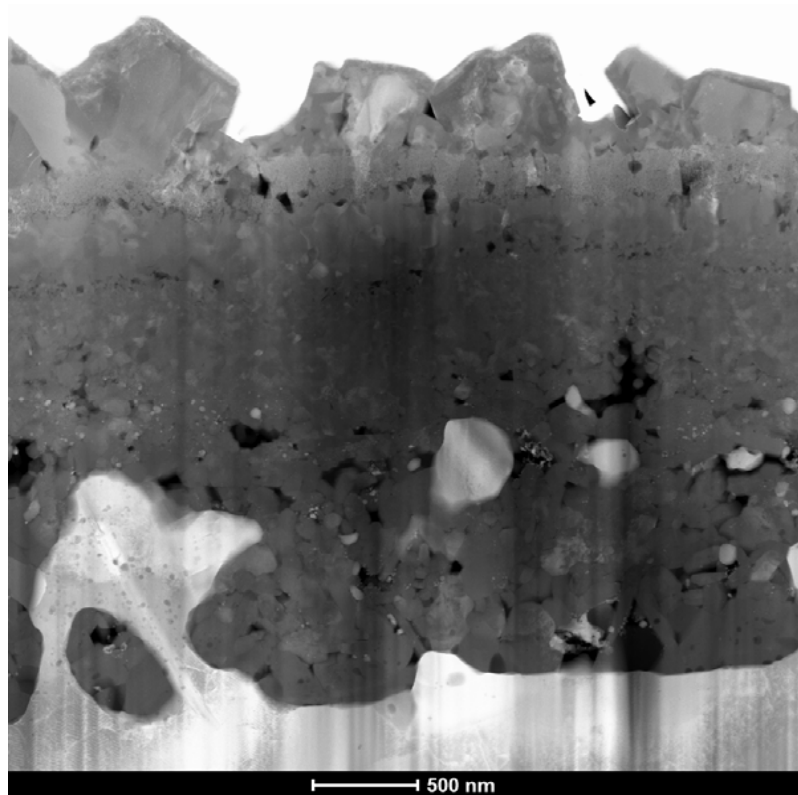


Metallography generally consistent with mass change data



625 (NiCr) 1000h STEM: unusually thick Cr-rich oxide layer

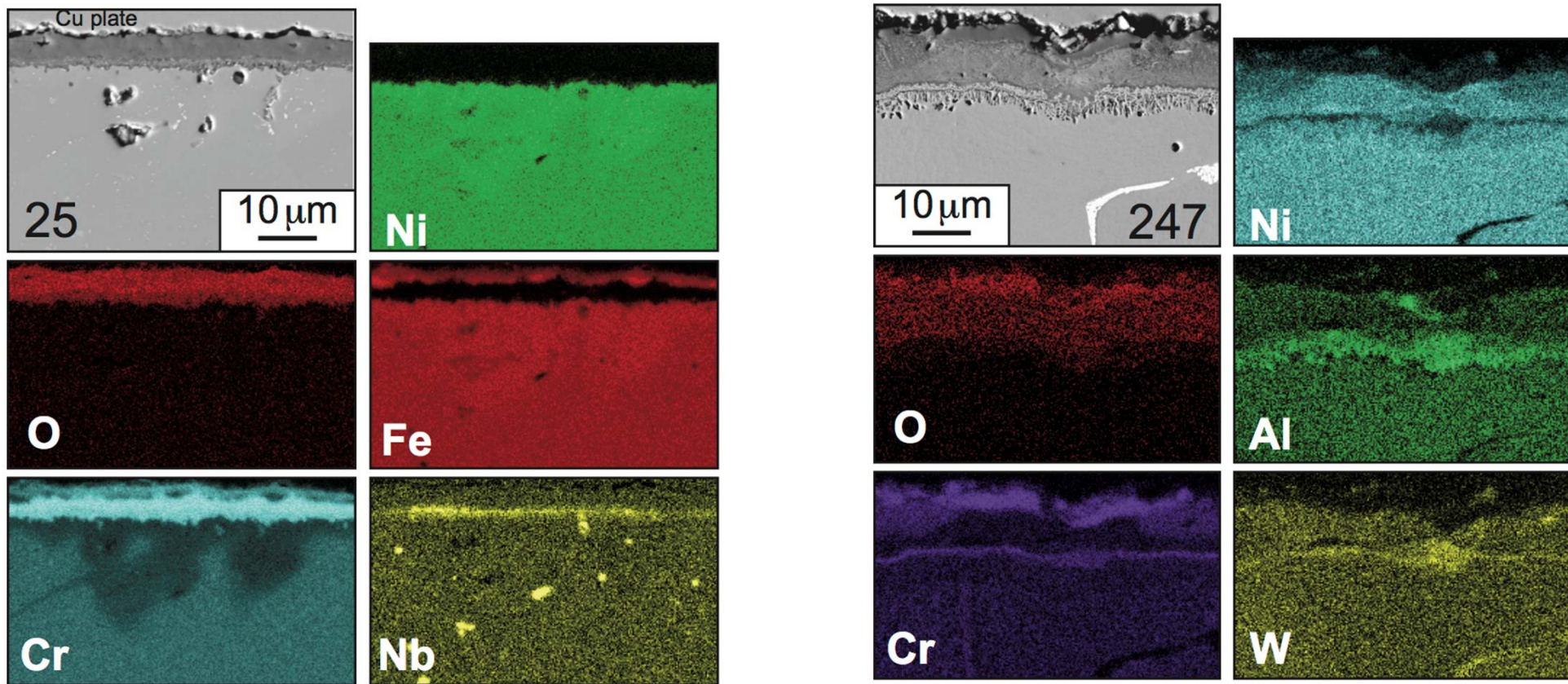
STEM annular dark field image



Scanning transmission electron microscopy Scale thickness $3.70 \pm 44 \mu\text{m}$

1000 h at 750°C , 300 bar $\text{sCO}_2 + 1\% \text{O}_2 + 0.25\% \text{H}_2\text{O}$

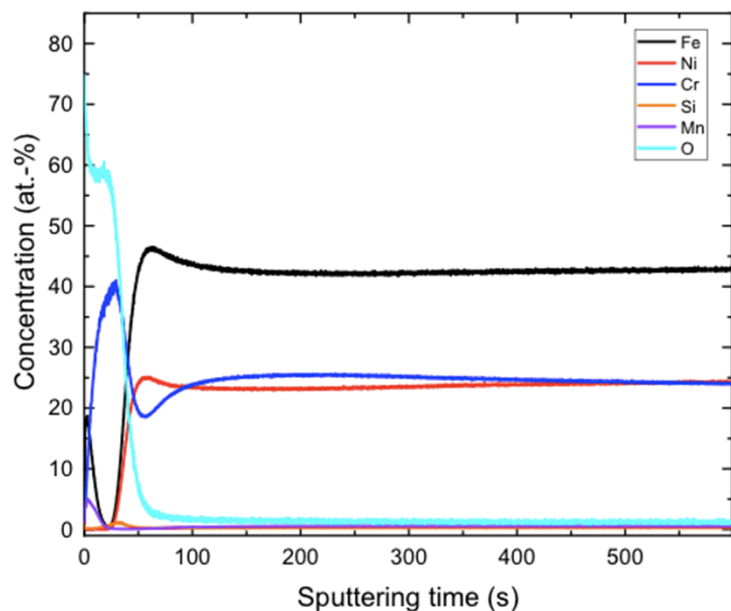
SEM/EDS: Fe/Ni-rich oxide forming with impurities



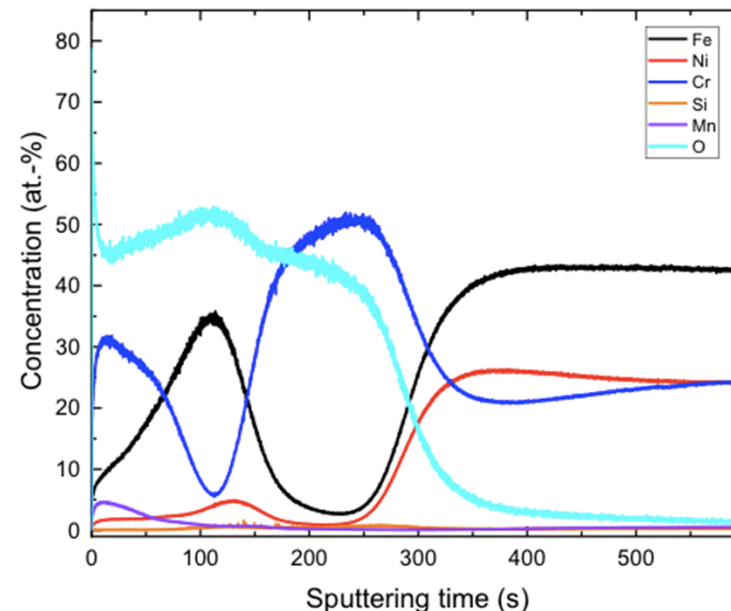
1000 h at 750°C, 300 bar $s\text{CO}_2 + 1\% \text{O}_2 + 0.25\% \text{H}_2\text{O}$

GDOES: very different oxide forming on 25: (Fe-22Cr-25Ni)

RG sCO₂



sCO₂+1%O₂+0.25%H₂O



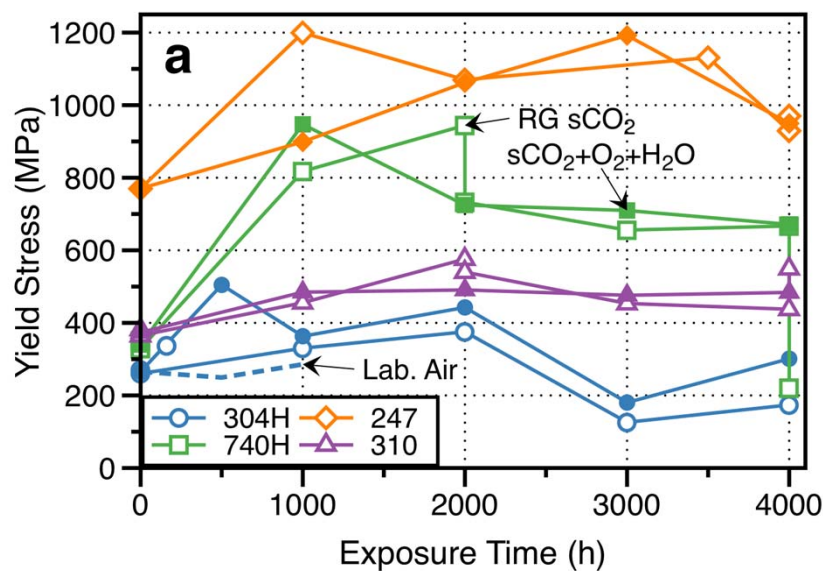
1000 h at 750°C, 300 bar

Glow discharge optical emission spectroscopy

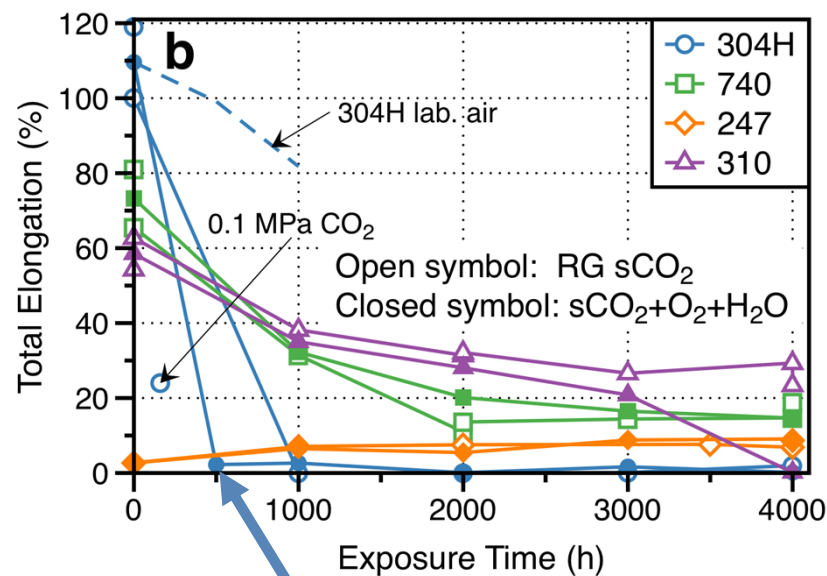
Much more work to understand role of O₂, H₂O on sCO₂ compatibility

750°C materials available, what about lower temperatures? Measuring 25°C tensile properties suggests future strategy: Method to quantify degradation of steels at 450°-650°C (?)

Yield strength



Elongation:



304H: large drop in sCO₂

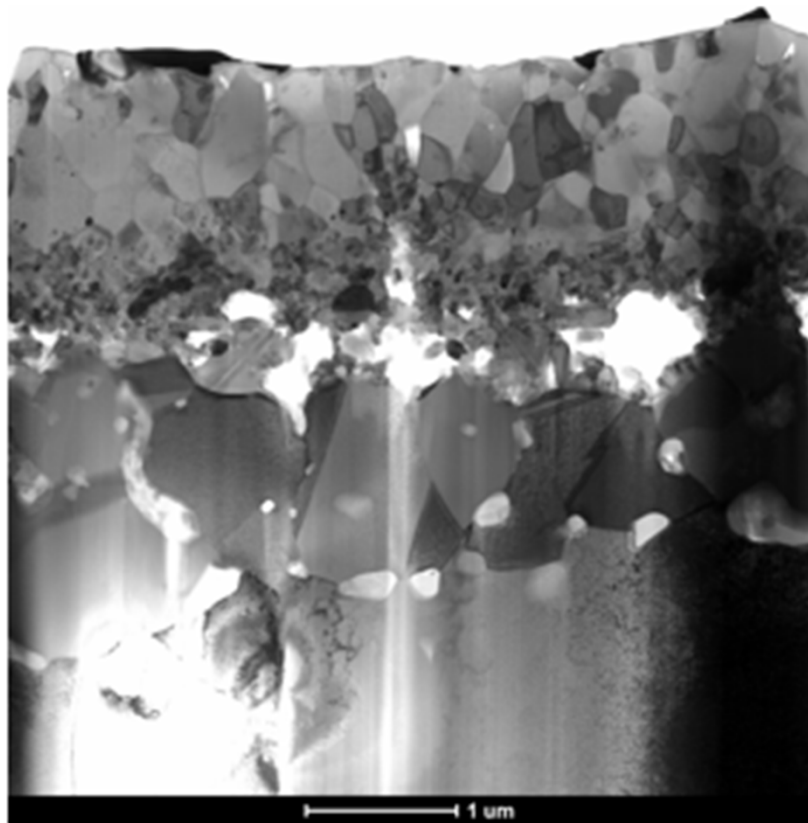
310: also degrading, w/o impurities

Pint, et al., Materials and Corrosion 70 (2019) 1400

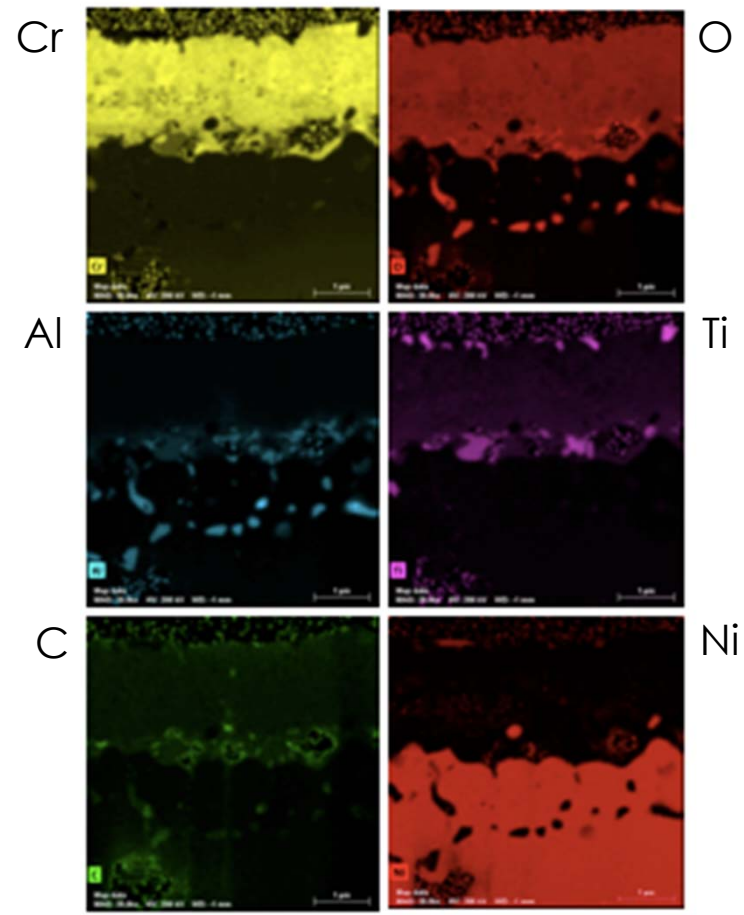
Backups

More complicated scale formed on alloy 282 specimen
1,000 h at 750°C in 300 bar sCO₂

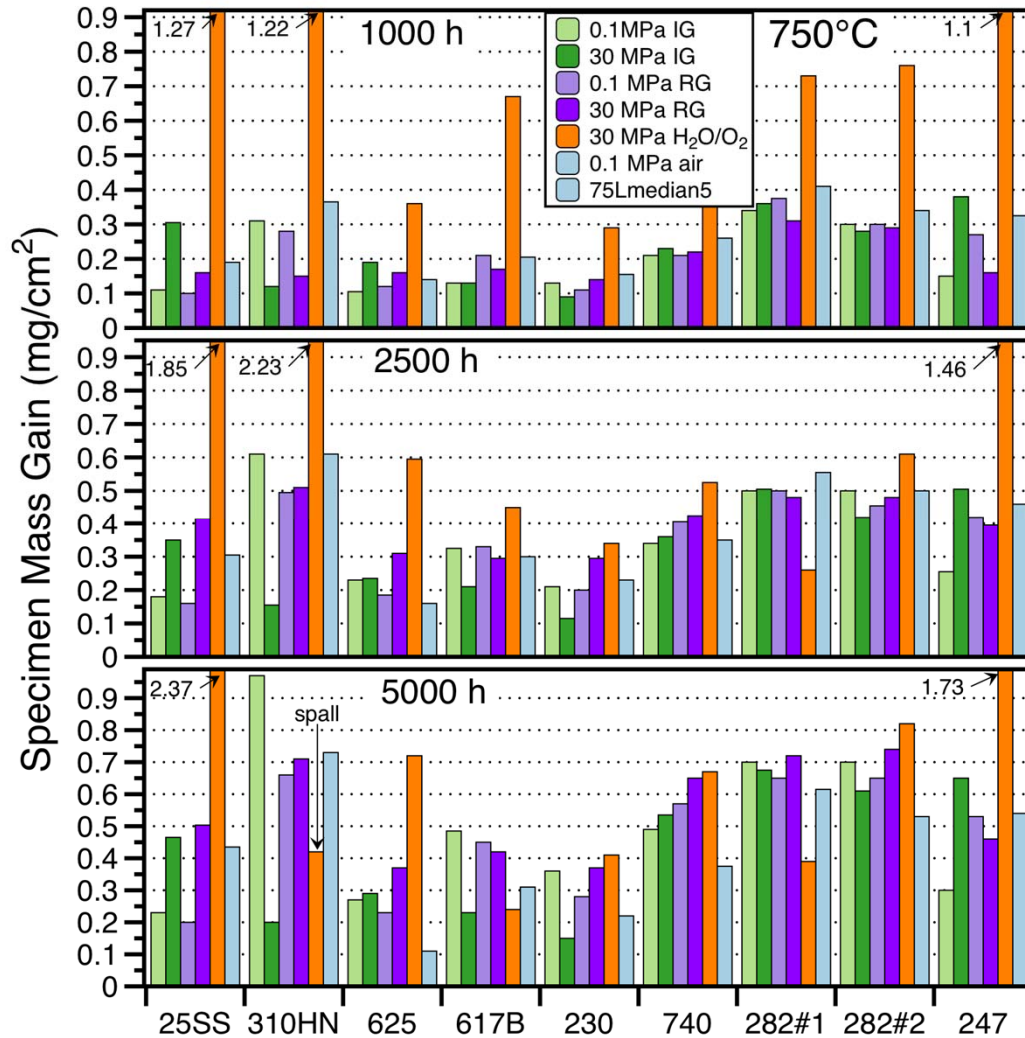
STEM BF image



X-ray maps

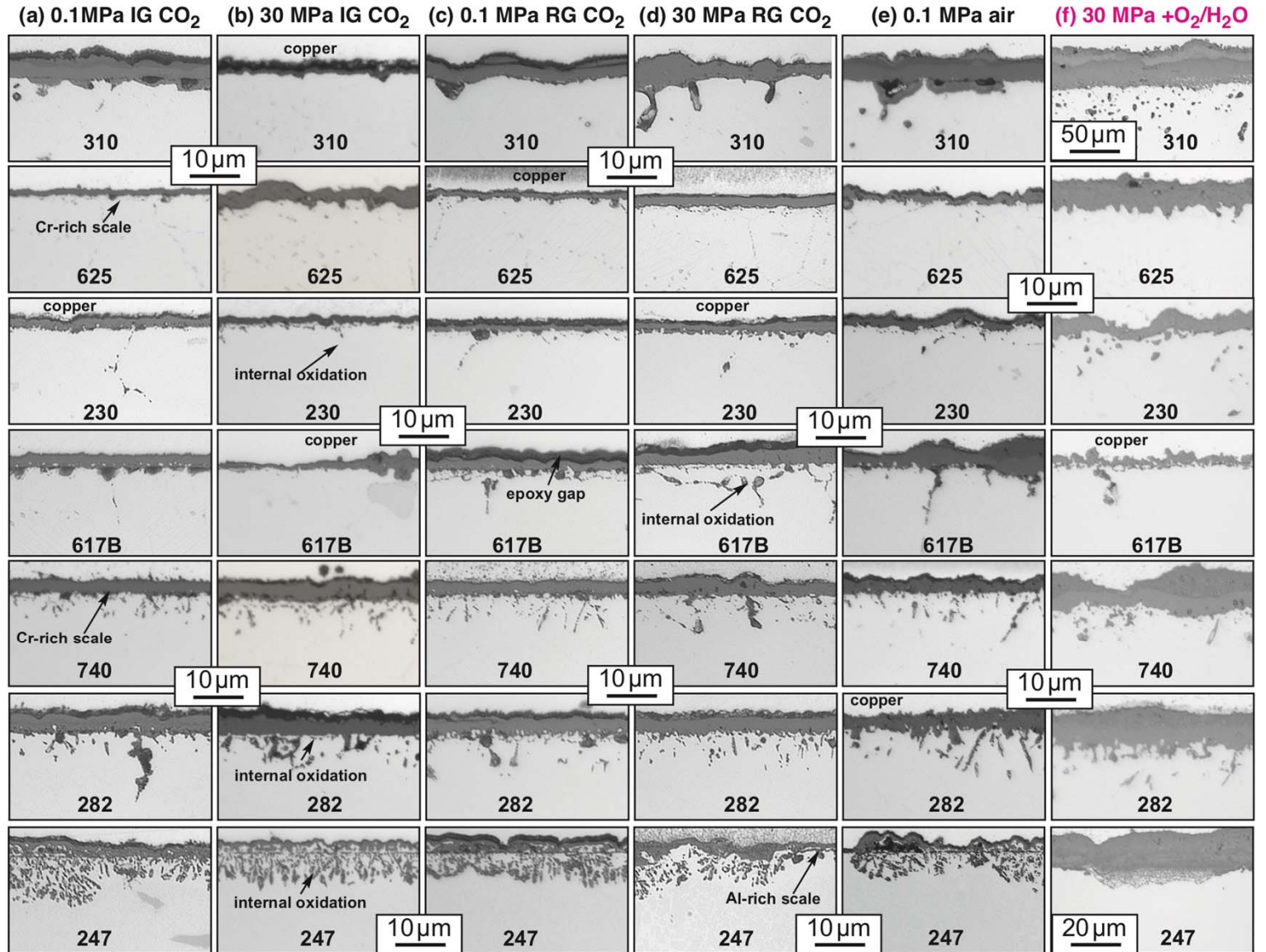
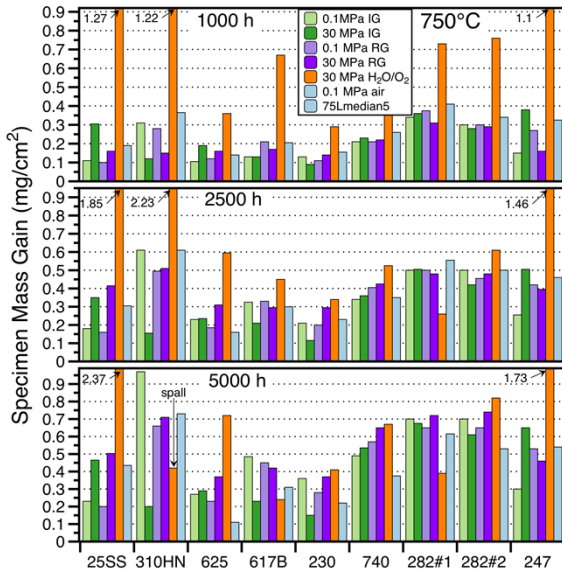


Summary: impurities caused a higher mass gain



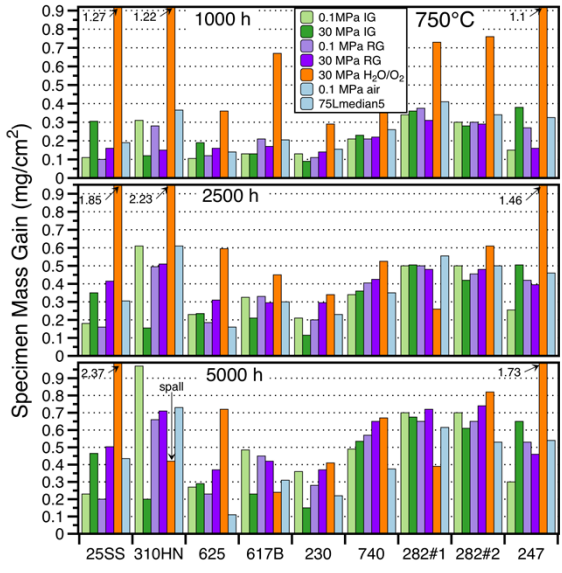
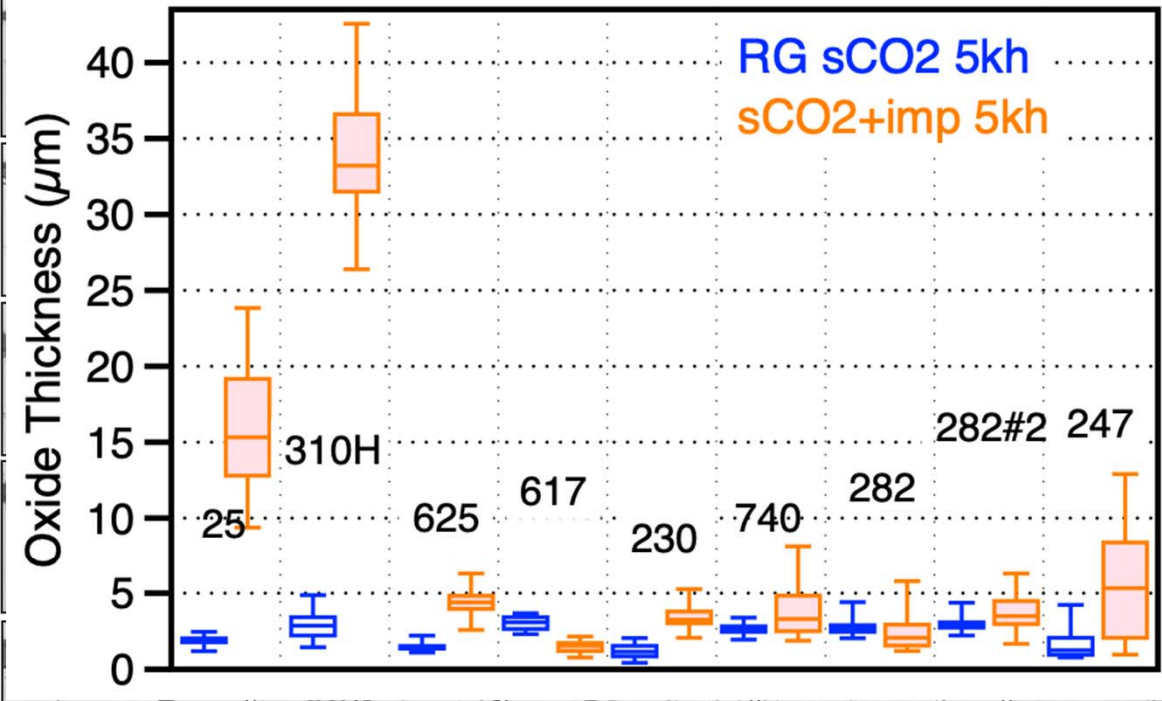
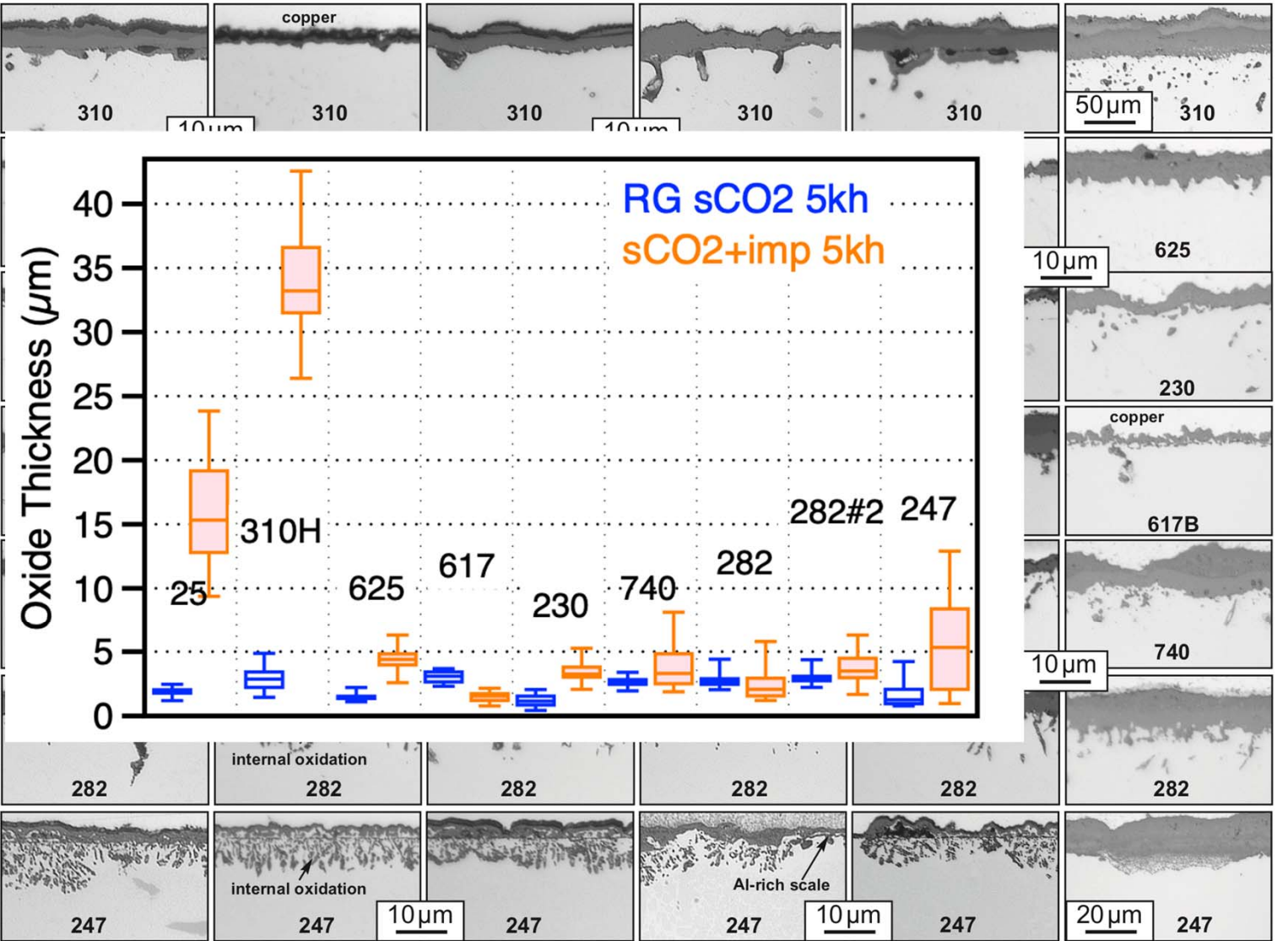
RG CO₂ + 1%O₂ + 0.25% H₂O

Metallography consistent with mass change



Fe-based alloys: strongest effect

(a) 0.1MPa IG CO₂ (b) 30 MPa IG CO₂ (c) 0.1 MPa RG CO₂ (d) 30 MPa RG CO₂ (e) 0.1 MPa air (f) 30 MPa +O₂/H₂O



Success warrants continued work to answer more q's

- Complete characterization and publications in early 2019
 - Mount 5,000 h specimens, etching, TEM, EPMA, etc.
- New FWP to focus on:
 - Mechanistic understanding of O₂/H₂O effects with ¹⁸O/H₂¹⁸O tracers
 - Separate O₂ and H₂O effects
 - **Focus on steels at lower temperatures: technology enabler**
 - #1 question from industry is "where can I use T91?": cheaper and strong vs. fear
 - Provide input to alloy designers on effect of Cr/Ni in Fe-Cr-Ni alloys
 - Use 25mm tensile bars to assess sCO₂ effect on strength & ductility
 - Surface modifications to improve steel performance (as warranted)
 - Shot peening
 - Coatings
 - Creep assessment of thin-walled material at 750°C (282/740/etc.)
 - Model lifetime as a function of temperature/HX dimensions/heat transfer

Project is concluding having achieved its goals

- Developed new experimental equipment for studying impurity effects on structural materials at up to 750°C/300 bar
- Results indicate that there is a negative effect of impurities, more so for Fe-based alloys than Ni-based alloys
- Room temperature tensile data suggests a simple route to quantify sCO₂ impact on mechanical properties of steels exposed at lower temperatures

Industrial collaborations

- All major US alloy manufacturers
- Conference calls conducted with HX, sCO₂ community (CSP project)
- Feedback from 8 Rivers/NetPower on impurity levels of interest
- Public presentations (TMS, MS&T, EFC workshop, NACE, EU sCO₂)