



Technische Universität Berlin

Institute for Energy Engineering



Study of a tri-generation system
based on a supercritical
CO₂ cycle

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Introduction

Research focuses on the application of exergy-based methods to improve the thermodynamic, economic and environmental performance of:

...

alternative power generation and refrigeration processes for sustainable industrial and commercial applications (**including CO₂ as working fluid**)

...

Exergy-Based Methods

Exergy-based methods is a general term that includes the conventional and advanced *exergetic*, *exergoeconomic*, and *exergoenvironmental analyses* and evaluations.

The *concept of exergy* complements and enhances an energetic analysis by calculating

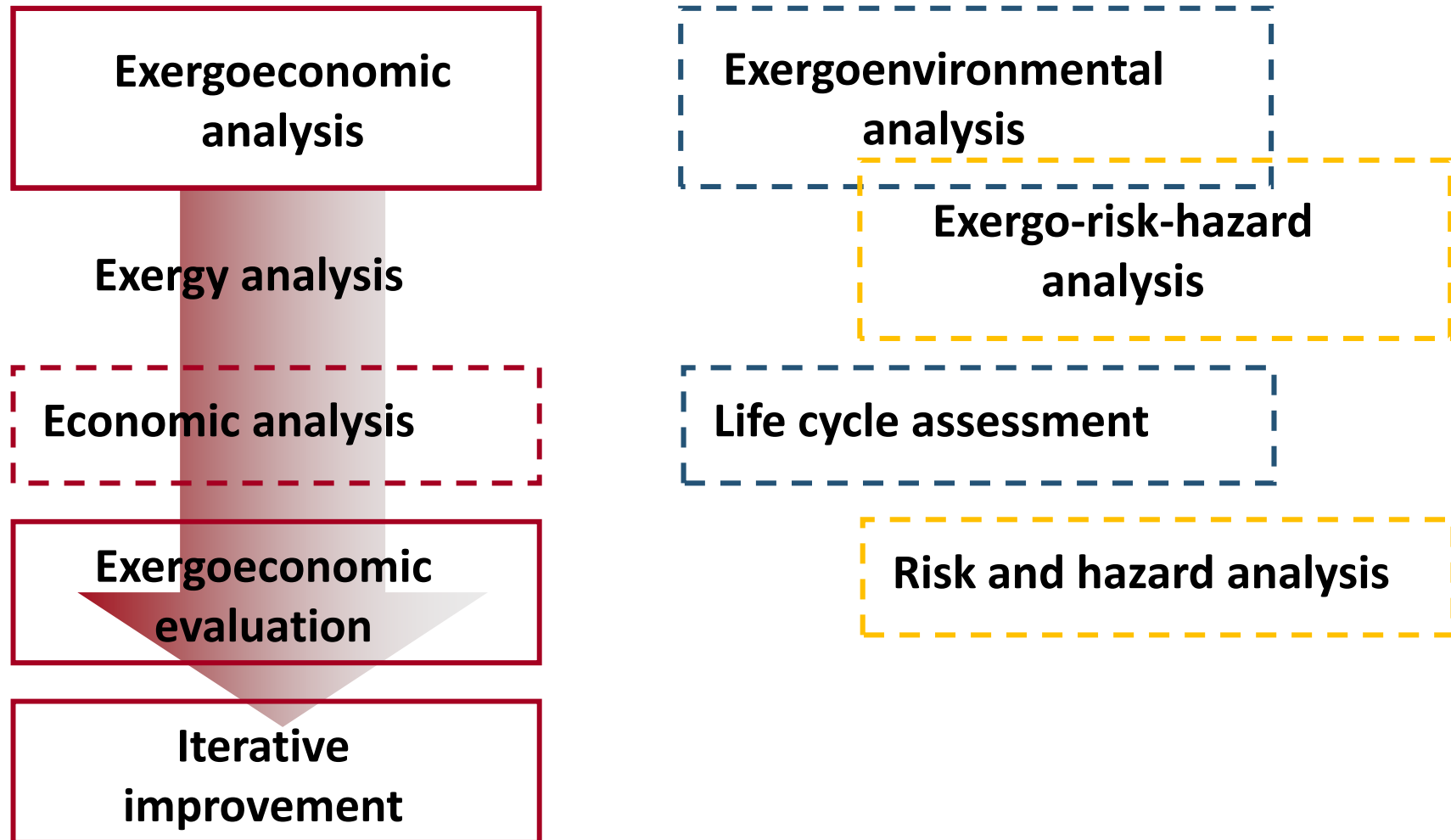
- (a) the true thermodynamic value of an energy carrier,
- (b) the real thermodynamic inefficiencies in a system, and
- (c) variables that unambiguously characterize the performance of a system (or one of its components) from the thermodynamic viewpoint.

Why Exergy-Based Methods? - 1

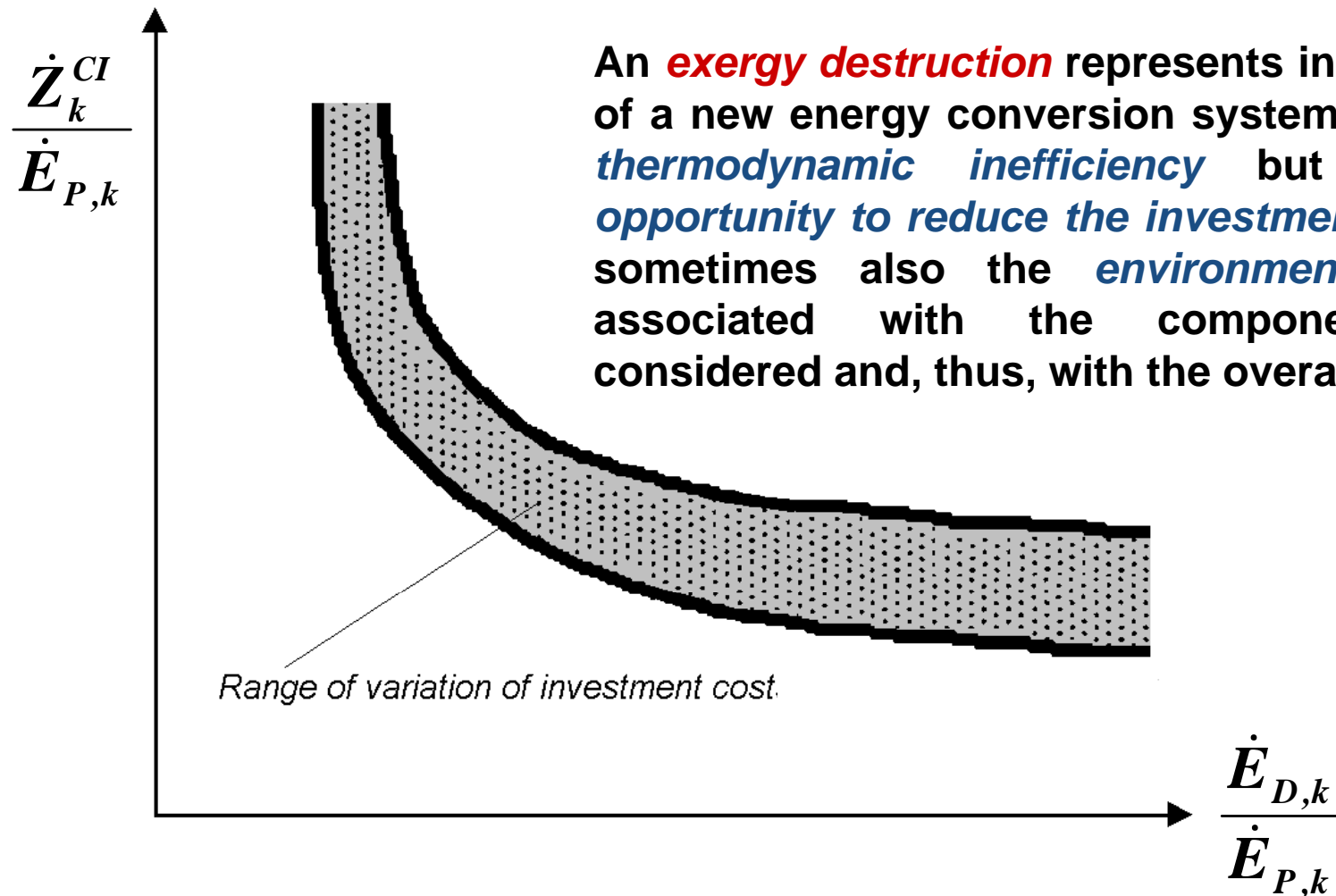
The objective evaluation and the improvement of an energy conversion system from the viewpoints of *thermodynamics*, *economics*, and *environmental impact* require a deep understanding of

- the *real thermodynamic inefficiencies* and the processes that cause them,
- the *costs associated with equipment and thermodynamic inefficiencies* as well as the connection between these two important factors,
- the *environmental impact associated with equipment and thermodynamic inefficiencies* as well as the connection between these two sources of environmental impact, and
- the *interconnections among efficiency, investment cost and component-related environmental impact* associated with the selection of specific system components

Exergy-based methods



Basic Principle of Exergoeconomics



Introduction – sCO₂

The idea of developing supercritical CO₂ power cycles and applying them to industrial processes became increasingly popular in the last decade.

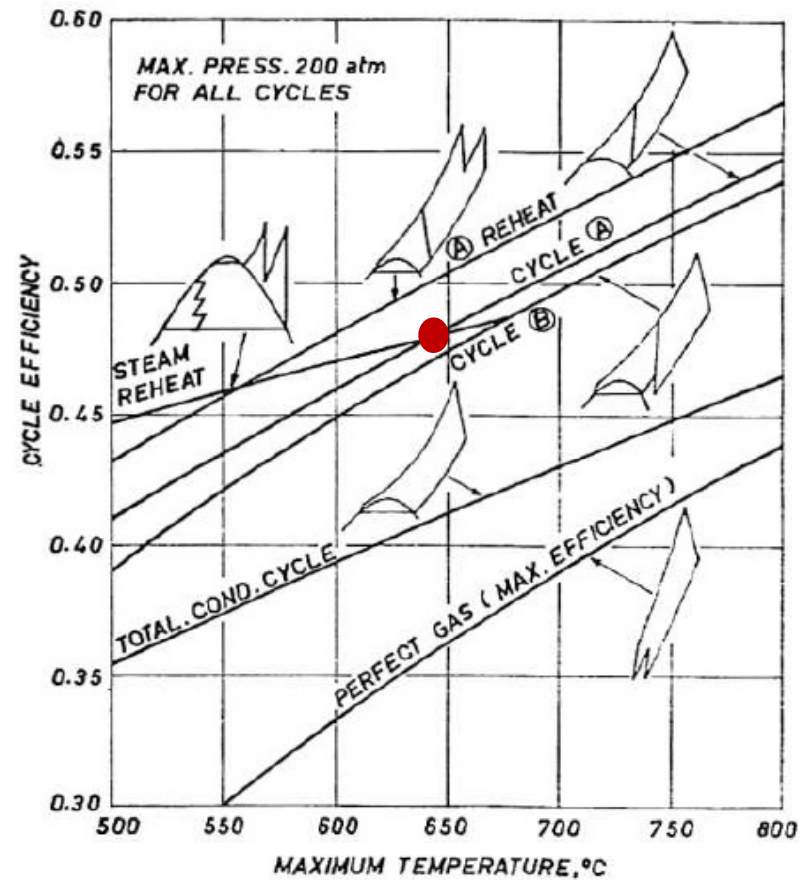
The potential for the application of supercritical CO₂ cycles is high for both *power generation systems* (Angelino, 1968) and *refrigeration systems* (Lorentzen, 1994) .

Significant research has been done in this field, including

- the development of new thermodynamic cycles,
- investigation of specific equipment, and
- parametric optimization of power systems.

Case Study – Power Cycles

$T_0 = 15^\circ\text{C}$



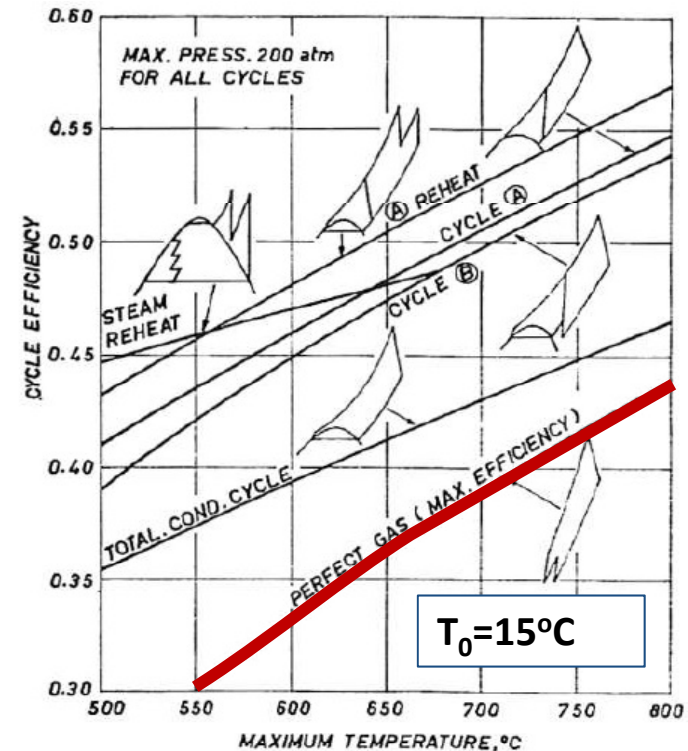
Angelino, 1968

New Challenges

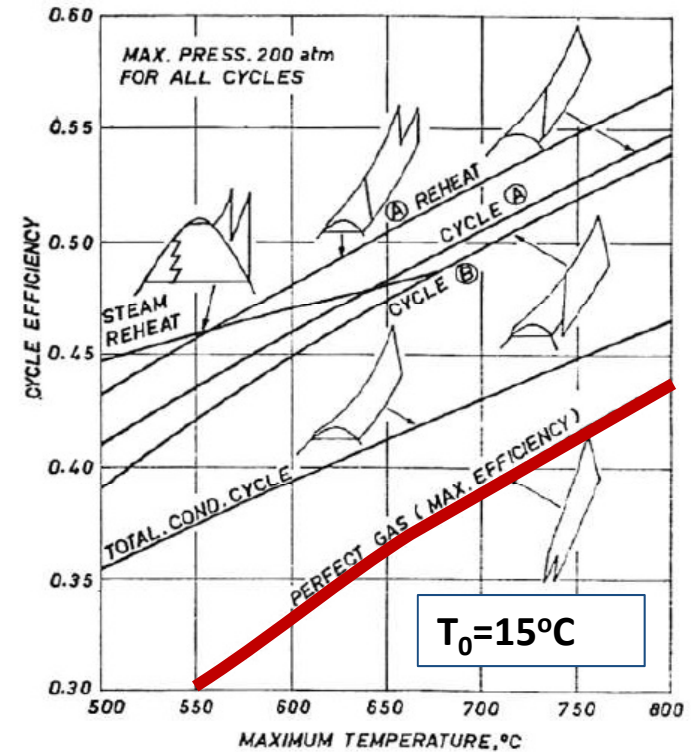
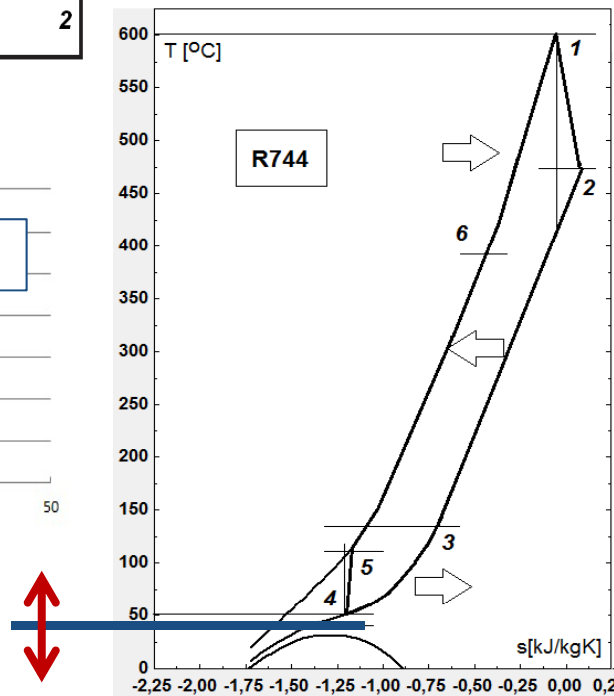
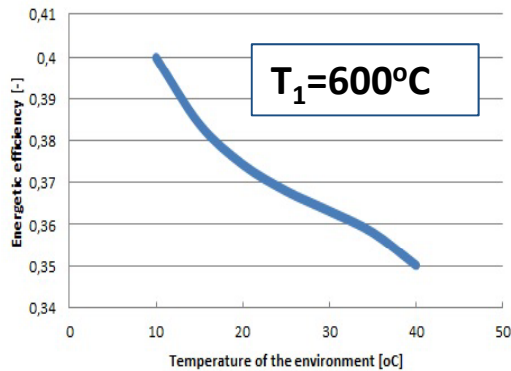
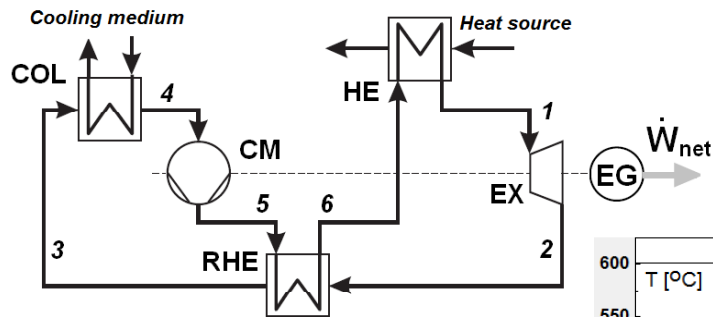
The present paper deals with the demonstration of the **application of advanced exergy-based methods** to supercritical CO₂ power cycles.

The examples used here assume power generation *when the temperature of the environment is high enough, so that a simple CO₂ cycle must operate above the critical temperature of CO₂ (i.e., >31.1 °C).*

Then none of the known condensation cycles can be applied.

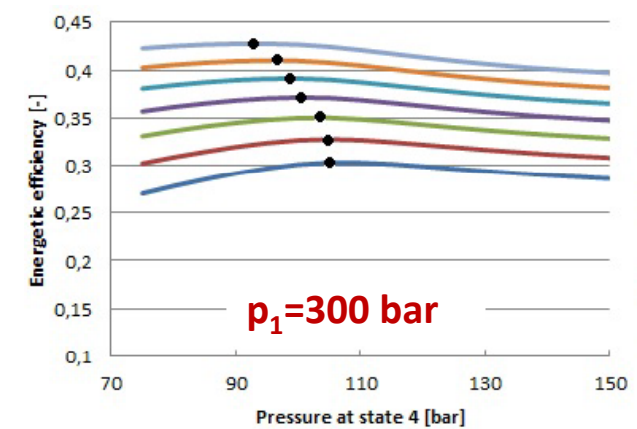
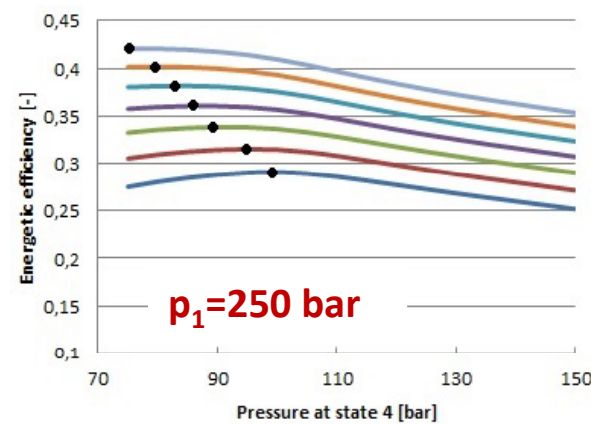
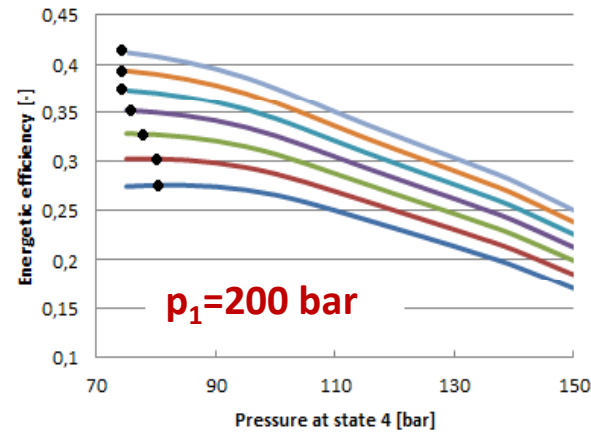
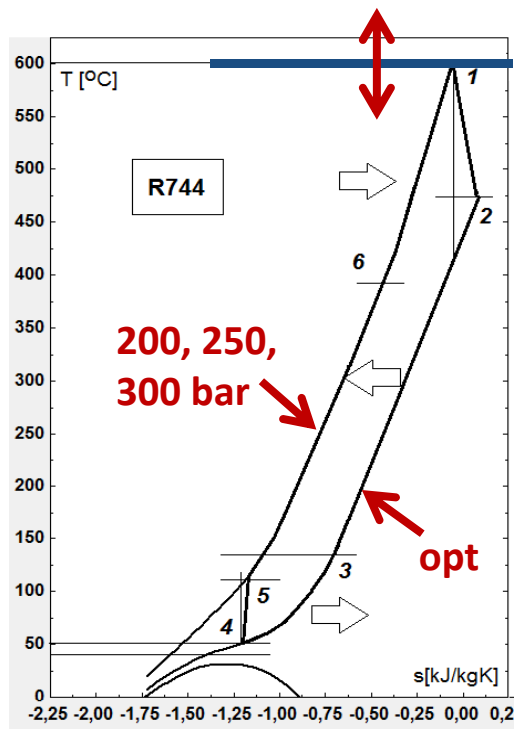


Case Study

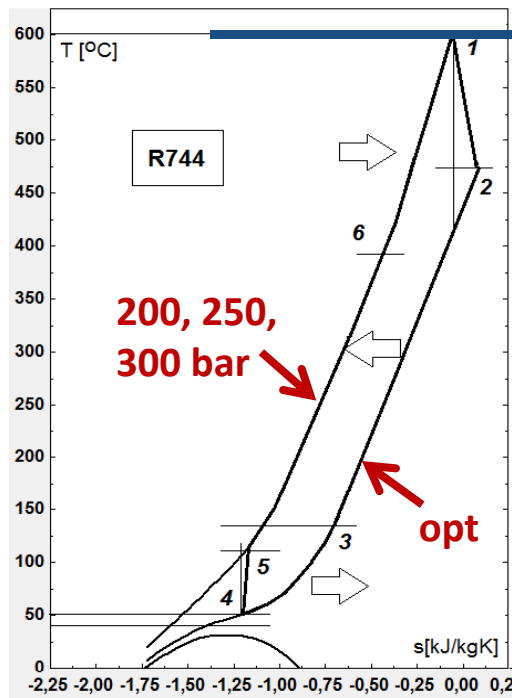


The temperature of the environment is high enough, so that a simple CO₂ cycle must operate above the critical temperature of CO₂ (i.e., >31.1 °C)

Case Study: Sensitivity Analysis



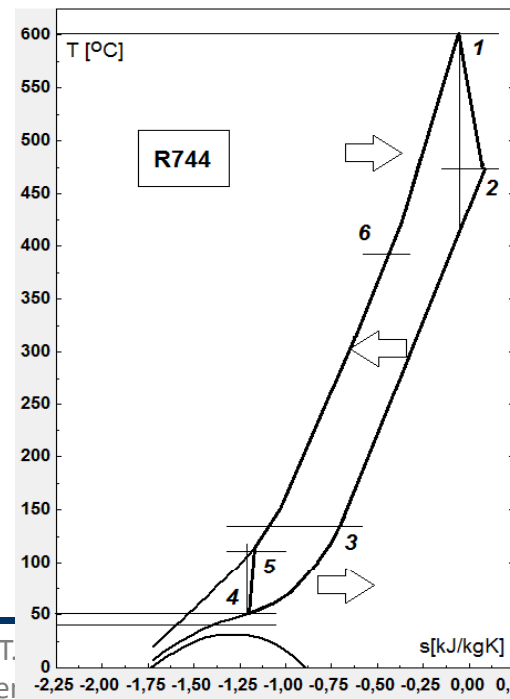
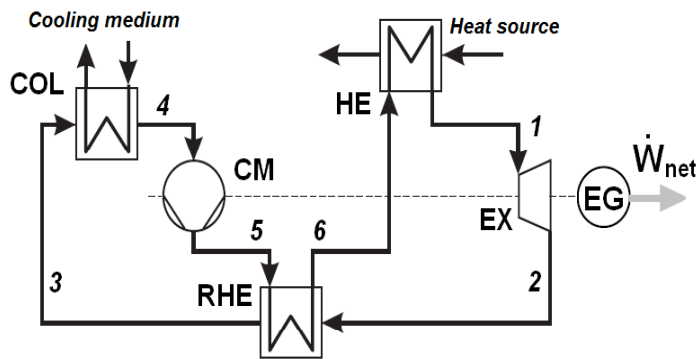
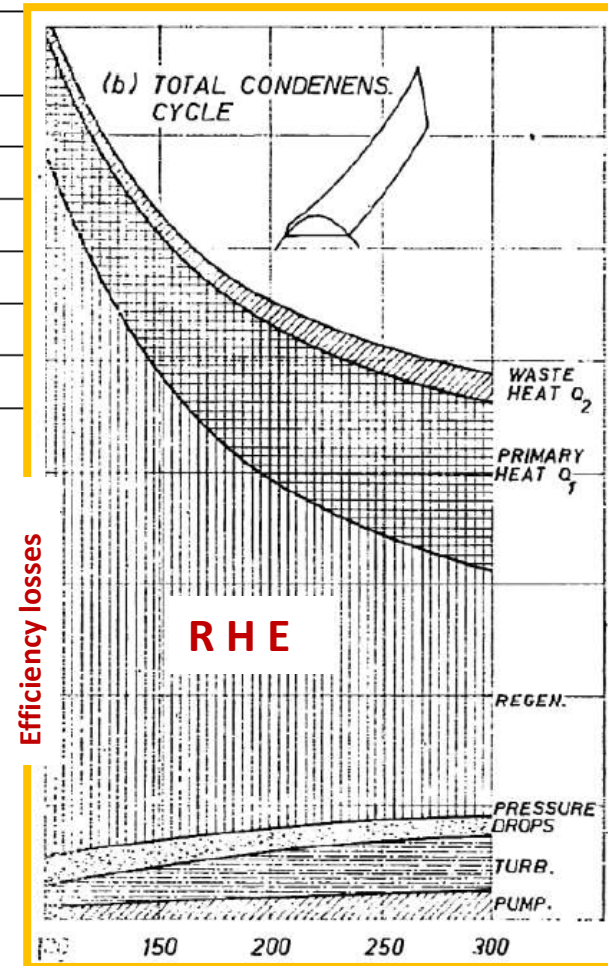
Case Study: Base Case



| p_1 [bar] | p_4^{opt} [bar] | η [-] | w_{net} [kJ/kg] | \dot{m}_{cycle} [kg/s] |
|----------------|----------------------|---------------|----------------------|-----------------------------|
| 200 | 75 | 0.33 | 73.15 | 136.7 |
| 250 | 90 | 0.34 | 83.95 | 119.1 |
| 300 | 105 | 0.35 | 94.97 | 105.3 |

Conventional Exergetic analysis: Results

| Component | $\dot{E}_{F,k}$ [kW] | $\dot{E}_{P,k}$ [kW] | $\dot{E}_{D,k}$ [kW] |
|----------------|----------------------|----------------------|----------------------|
| CM | 4,688 | 4,114 | 574 |
| COL | 2,215 | 1,249 | 966 |
| EX | 15,377 | 14,689 | 688 |
| HE | 16,939 | 16,623 | 316 |
| RHE | 19,036 | 15,891 | 3,145 |
| Overall system | 16,939 | 10,000 | 5,689 |

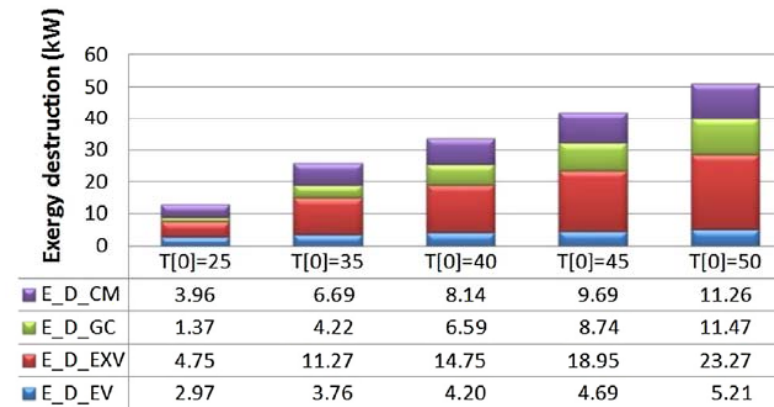
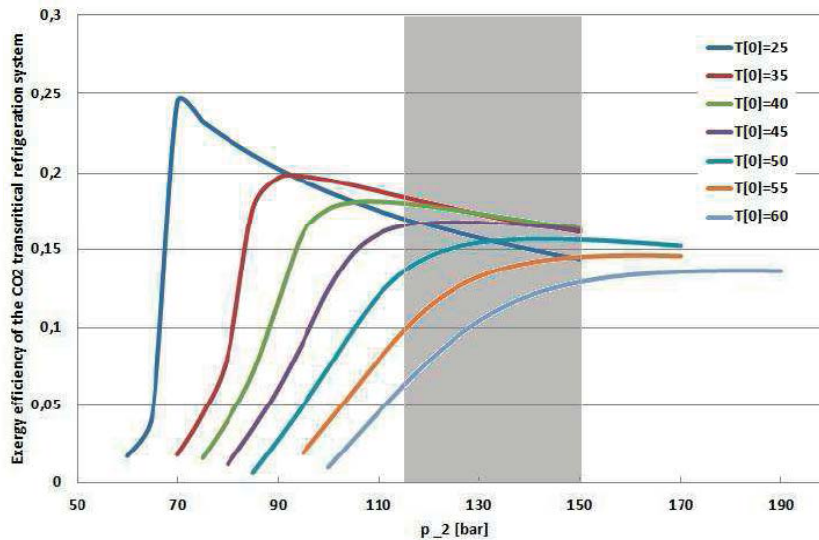
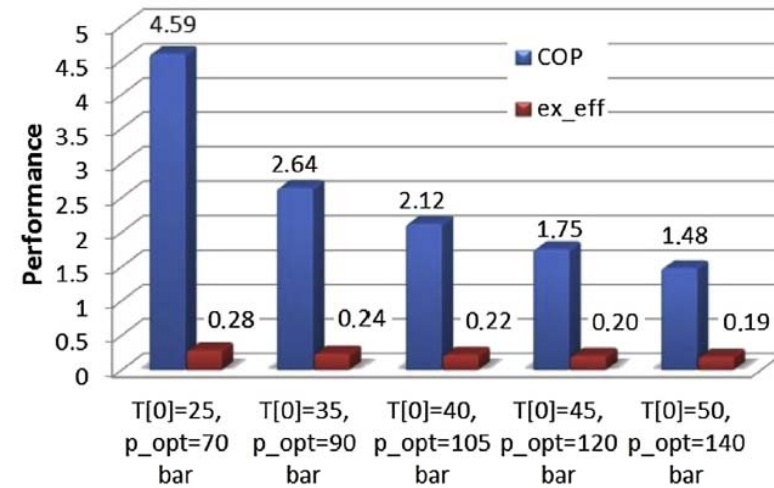
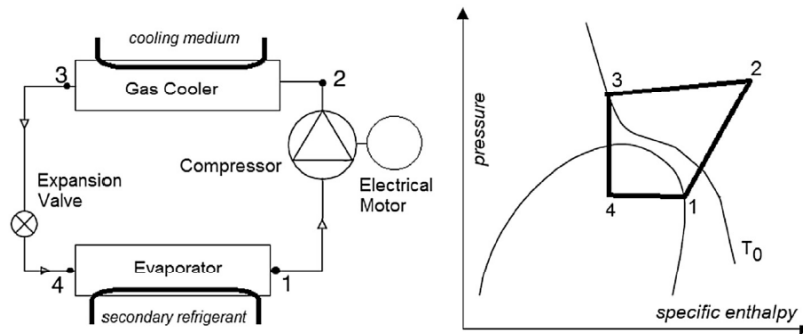


Conclusions – Power Cycles

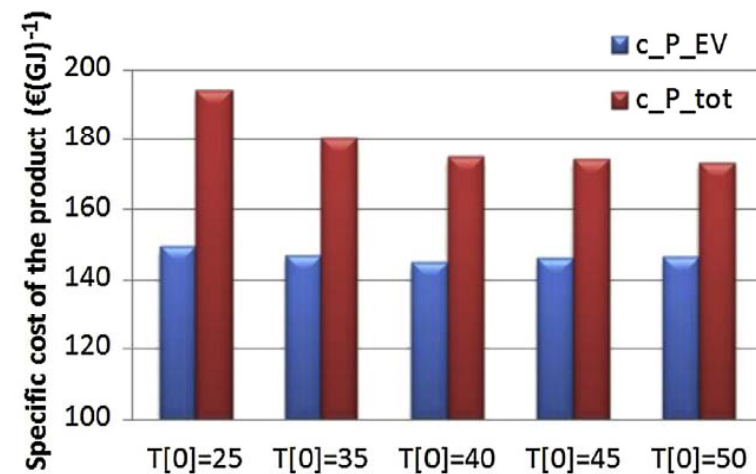
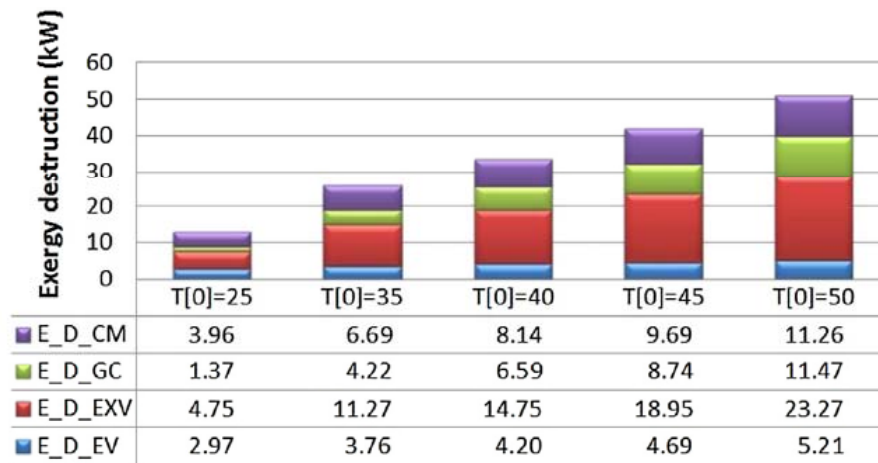
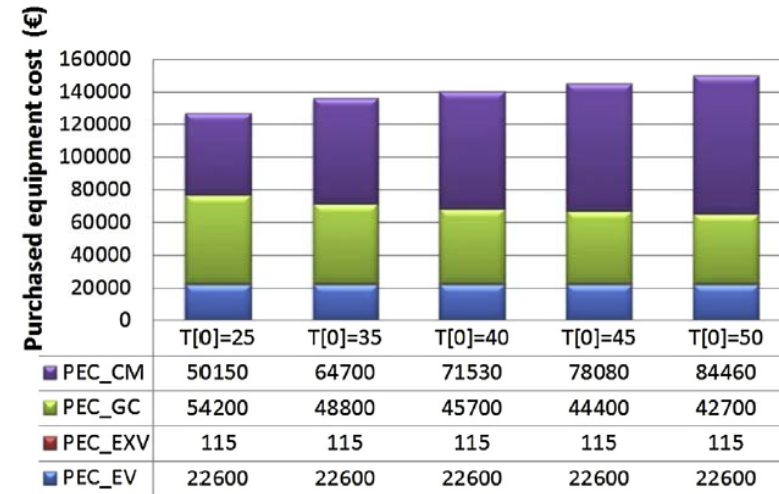
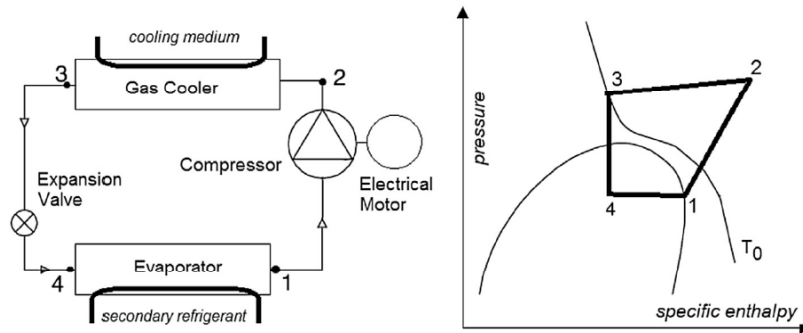
The application of an advanced exergetic analysis to a simple supercritical CO₂ cycle was demonstrated. This system can be improved by improving the components in isolation, because the *avoidable* inefficiencies caused by the components interconnections are relatively low.

The most important component from the thermodynamic viewpoint is the regenerative heat exchanger. System designers should focus on this component.

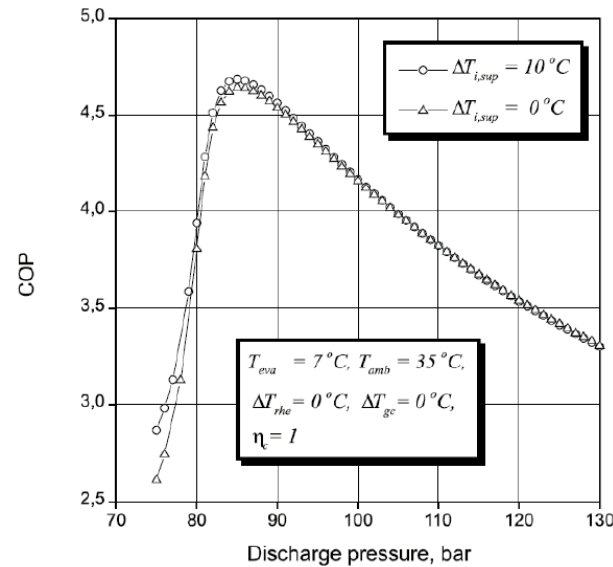
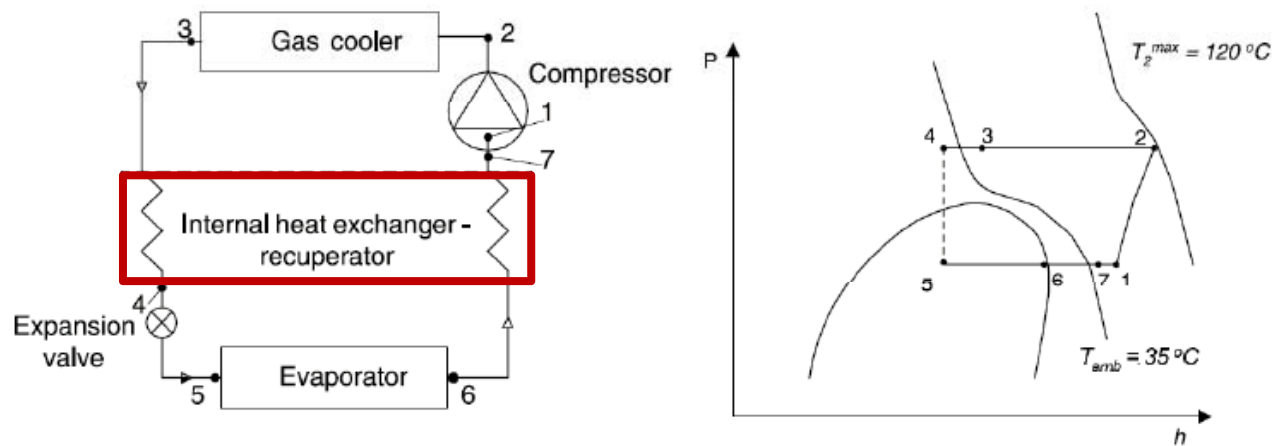
Refrigeration Applications – Exergy Analysis



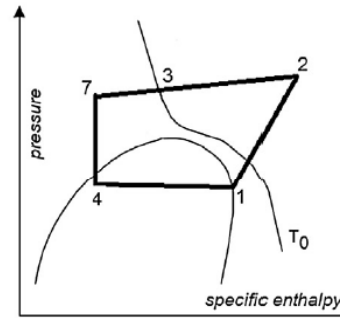
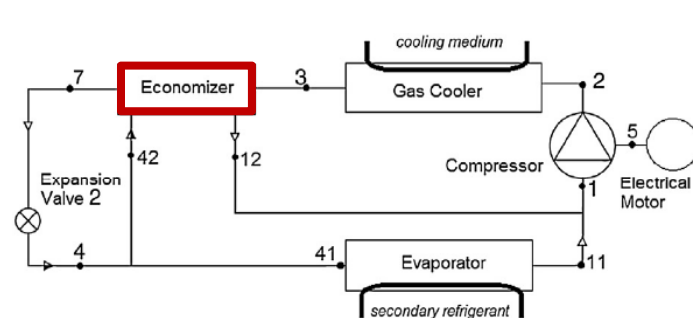
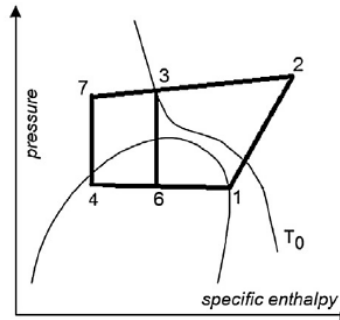
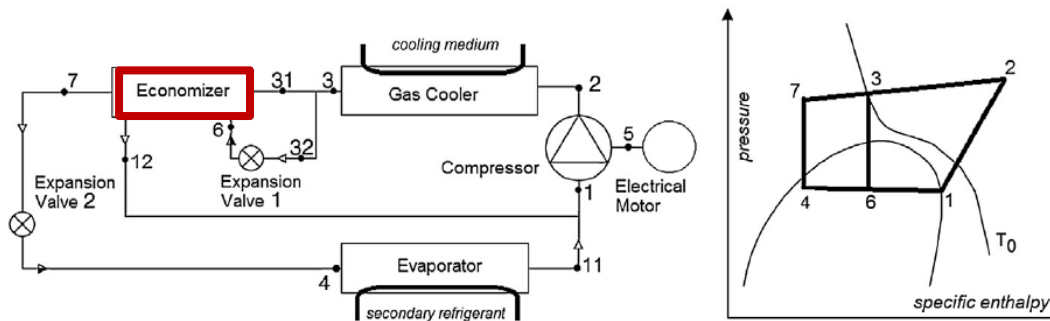
Refrigeration Applications – Cost Analysis



Refrigeration Applications – Improvements

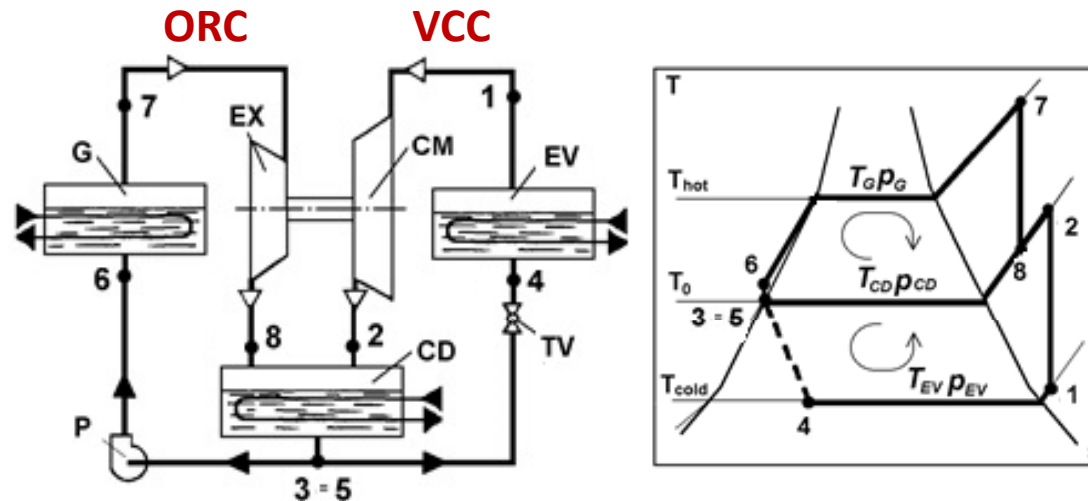


Refrigeration Applications – Improvements



The advantage of economizer was discovered through the exergy-based evaluation. Using both options, the *purchased equipment costs of the overall machine increases by 4-5%*, while the *total cost of the final product decreases for 13-14%*. Based on conducted evaluation we can conclude that for countries with hot climates, the one-stage sCO₂ refrigeration machine should include the economizer as an auxiliary component.

Refrigeration Applications – Heat-driven Machine

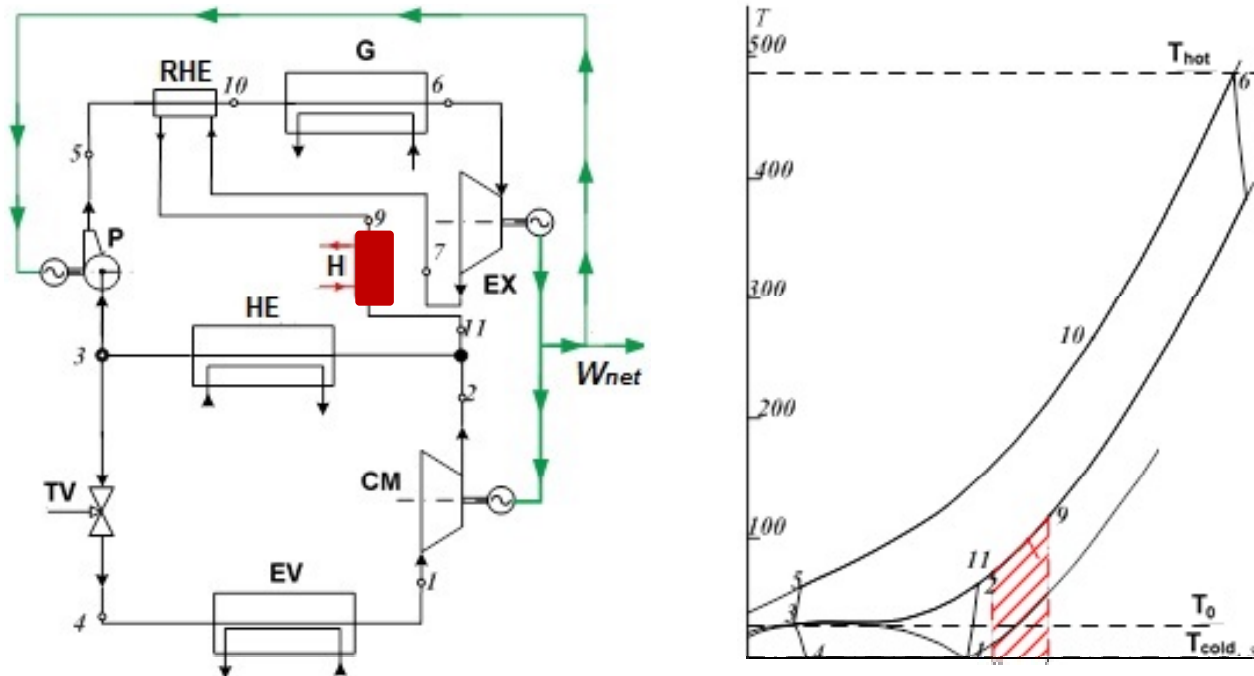


Chistiakov-Plotnikov
refrigeration machine,
Patent USSR 1952

In the concept to be presented here deals with the extended idea of the Chistiakov-Plotnikov refrigeration machine that includes the following:

- CO_2 is the working fluid,
- VCC is replaced with a Supercritical Refrigeration Cycle (SRC), and
- ORC is replaced with a closed-cycle gas turbine system.

Refrigeration Applications – Heat-driven Machine



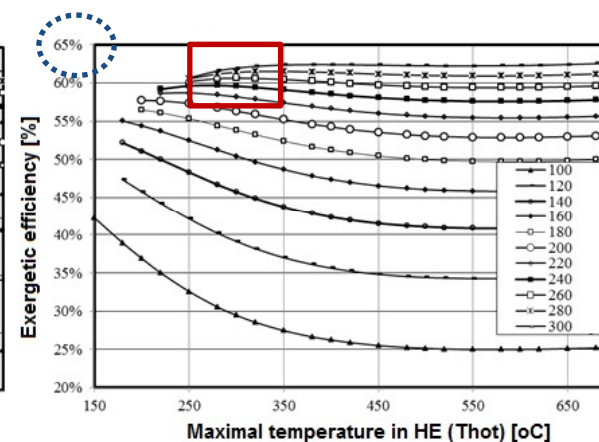
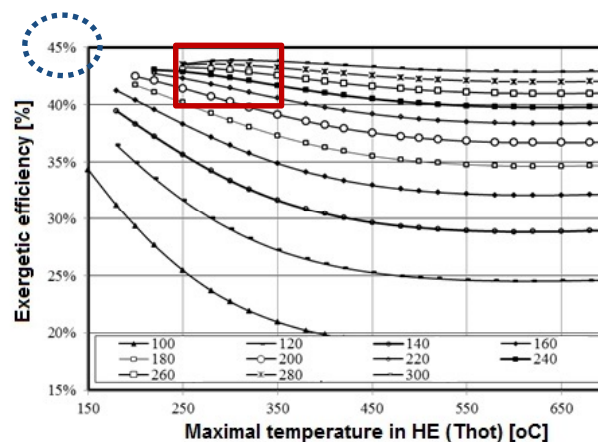
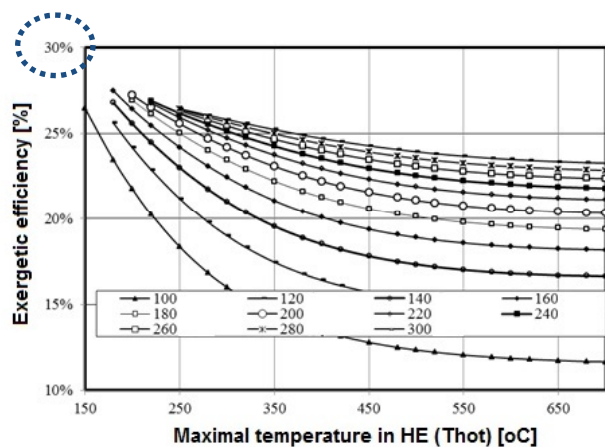
- The excess power \dot{W}_{net} is generated because $\dot{W}_{direct\ cycle} > \dot{W}_{ref\ cycle}$;

\dot{W}_{net} is the first product stream (positive effect), and

- The heat produced within the heater (H) represents the second thermal product (effect) of the machine, in addition to the refrigeration capacity

Refrigeration Applications – Heat-driven Machine

The values of \dot{Q}_{cold} and \dot{W}_{net} depend on the maximal temperature within the generator (T_{hot}) and pressure, but the ratio $z = \dot{W}_{net} / \dot{Q}_{cold}$ can be easily varied depends on the demand.



Conclusions

A novel tri-generation concept of a combined cycle coupled with ORC and SRC with R744 as working fluid has been introduced and evaluated using exergy analysis.

Future investigations will include the evaluation of more complex configurations and the application of economic and exergoeconomic analyses in order to identify the configurations and the system parameters that will lead to a lower cost of the generated products.

Thank you
for your attention

Exergetic Variables: E_P and E_F

Exergy of product: \dot{E}_P

The desired result, expressed in exergy terms, achieved by the system (the k -th component) being considered.

Exergy of fuel: \dot{E}_F

The exergetic resources expended to generate the exergy of the product.

The concepts of product and fuel are used in a consistent way not only in *exergetic analyses* but also in the *exergoeconomic* and *exergoenvironmental* analyses.

Exergetic Variables: E_D and E_L

Exergy destruction: \dot{E}_D

Exergy destroyed due to irreversibilities within a system (the k -th component).

Exergy loss: \dot{E}_L

Exergy transfer to the system surroundings. This exergy transfer is not further used in the installation being considered or in another one.

Exergy balance:

$$\dot{E}_F = \dot{E}_P + \dot{E}_D (+ \dot{E}_L)$$

\dot{E}_D and \dot{E}_L are **absolute** measures of the thermodynamic inefficiencies.

Exergetic Variables: ε and $y_{D,k}$

Relative measures of the thermodynamic inefficiencies:

Exergetic efficiency: The ratio between exergy of product and exergy of fuel

$$\varepsilon_k = \frac{\dot{E}_{P,k}}{\dot{E}_{F,k}}$$

Exergy destruction ratio for the k -th component

$$y_{D,k} = \frac{\dot{E}_{D,k}}{\dot{E}_{F,tot}}$$